

TITLE OF THE INVENTION

RESIST COMPOSITION

BACKGROUND OF THE INVENTION

1. Field of the Invention

5 The present invention relates to a novel, chemically amplified resist composition suited for use in micro-lithography.

2. Description of the Related Art

The miniaturization of a pattern rule has been demanded 10 in order to cope with a recent tendency of LSI technology to higher integration and higher speed. Under such a circumstance, far ultraviolet lithography has been regarded promising as the next-generation of micro-lithography. Even 15 a pattern of 0.3  $\mu\text{m}$  or less can be formed by far ultraviolet lithography, and use of a resist material exhibiting low light absorption makes it possible to form a pattern with side-wall angles nearly vertical to a substrate. In recent years, a technique making use of a high intensity KrF excimer 20 laser as a far-UV light source has drawn attentions. There is accordingly a demand for the development of a low light absorbing and highly sensitive resist material which permits the use of the above-described technique for mass production.

From such a viewpoint, the recently-developed, 25 chemically amplified positive type resist materials as described in Japanese Patent Publication (JP-B) No. 2-

27660/'90 and Japanese Patent Provisional Publication (JP-A) No. 63-27829/'88 using an acid catalyst are particularly promising resist materials suited for far-UV lithography, because of their excellent characteristics such as high 5 sensitivity, resolution and dry etching resistance.

Prior-art chemically amplified resists are however accompanied with the problems such as PED (Post Exposure Delay) that line patterns have a T-top profile, in other words, patterns become thick at the top, when the dwelling 10 time from exposure to PEB (post exposure bake) is extended; and a so-called "trailing phenomenon" that patterns in the vicinity of a basic substrate, particularly, a substrate made 15 of silicon nitride or titanium nitride are widened.

It is presumed that the T-top profile results from lowering in the solubility of the surface of a resist film, while the trailing phenomenon on a substrate results from 20 lowering in solubility in the vicinity of the substrate.

In addition, during the time from exposure to PEB, a dark reaction for eliminating an acid-labile group proceeds, leading to a problem that the leaving size of the line of a 25 positive resist decreases.

These problems are serious hindrance to practical use of the chemically amplified resist. Such problems of the conventional chemically amplified positive resist material 25 not only make difficult dimensional control upon lithography

but also impair dimensional control upon processing of a substrate by using dry etching [refer to: W. Hinsberg, et al., J. Photopolym. Sci. Technol., 6(4), 535-546(1993) and T. Kumada, et al., J. Photopolym. Sci. Technol., 6(4), 571-574(1993)].

It is understood that in these chemically amplified positive type resist materials, a basic compound in the air or on the surface of a substrate largely takes part in the problem of PED or trailing phenomenon on the substrate. An acid on the surface of a resist film generated by exposure to light reacts with a basic compound in the air and is thereby deactivated. The longer the dwelling time from exposure to PEB becomes, the more the amount of a deactivated acid increases, making it difficult to cause decomposition of an acid-labile group. An insolubilized layer is therefore formed on the surface and a pattern inevitably has a T-top configuration.

It is well known that addition of a basic compound is effective for overcoming PED, because it can suppress the influence of a basic compound in the air (as described in Japanese Patent Provisional Publication Nos. 5-232706/'93, 5-249683/'93, 5-158239/'93, 5-249662/'93, 5-257282/'93, 5-289322/'93, 5-289340/'93, 6-194834/'93, 6-242605/'94, 6-242606/'94, 6-263716/'94, 6-263717/'94, 6-266100/'94, 6-266111/'94, 7-128859/'95, 7-92678/'95, 7-92680/'95, 7-

92681/'95, 7-120929/'95 and 7-134419/'95).

As the basic compound, a nitrogen-containing compound is well known and examples include amine compounds and amide compounds having a boiling point of 150°C or greater.

5 Specific examples include pyridine, polyvinyl pyridine, aniline, N-methylaniline, N,N-dimethylaniline, o-toluidine, m-toluidine, p-toluidine, 2,4-lutidine, quinoline, isoquinoline, formamide, N-methylformamide, N,N-dimethylformamide, acetamide, N-methylacetamide, N,N-dimethylacetamide, 2-pyrrolidone, N-methylpyrrolidone, imidazole,  $\alpha$ -picoline,  $\beta$ -picoline,  $\gamma$ -picoline, o-aminobenzoic acid, m-aminobenzoic acid, p-aminobenzoic acid, 1,2-phenylenediamine, 1,3-phenylenediamine, 1,4-phenylenediamine, 2-quinolincarboxylic acid, 2-amino-4-nitrophenol, and 15 triazine compounds such as 2-(p-chlorophenyl)-4,6-trichloromethyl-s-triazine. Among them, pyrrolidone, N-methylpyrrolidone, o-aminobenzoic acid, m-aminobenzoic acid, p-aminobenzoic acid and 1,2-phenylenediamine are typical examples.

20 SUMMARY OF THE INVENTION

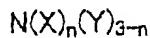
Although the above-described nitrogen-containing compounds can ease the T-top problem at an acid dissociation constant pKa ranging from 2 to 6, they cannot control reaction, that is, acid diffusion upon use of a highly-reactive acid-labile group.

When a weak base is added, a dark reaction in PED proceeds at an unexposed portion, thereby causing a reduction in a line size (slimming) and a decrease in film thickness on the line surface. Addition of a strong base having a pKa of 5 7 or greater is effective for overcoming the above-described problem.

However, higher pKa does not always bring about good results. Even when a superstrong base such as DBU (1,8-diazabicyclo[5.4.0]-7-undecene) or DBN (1,5-diazabicyclo[4.3.0]-5-nonen), proton sponge or a quaternary amine such as tetramethylammonium hydroxide is added, a sufficient effect is not available.

The present inventors have carried out various investigations. As a result, it has been found that amines represented by formulas (I) to (III) and (1) to (4) having a carbonyl group, an ester group, or a carbonate group are highly effective for preventing a decrease in the thickness of a resist film and also for enlarging a focus margin and bringing about dimensional stability.

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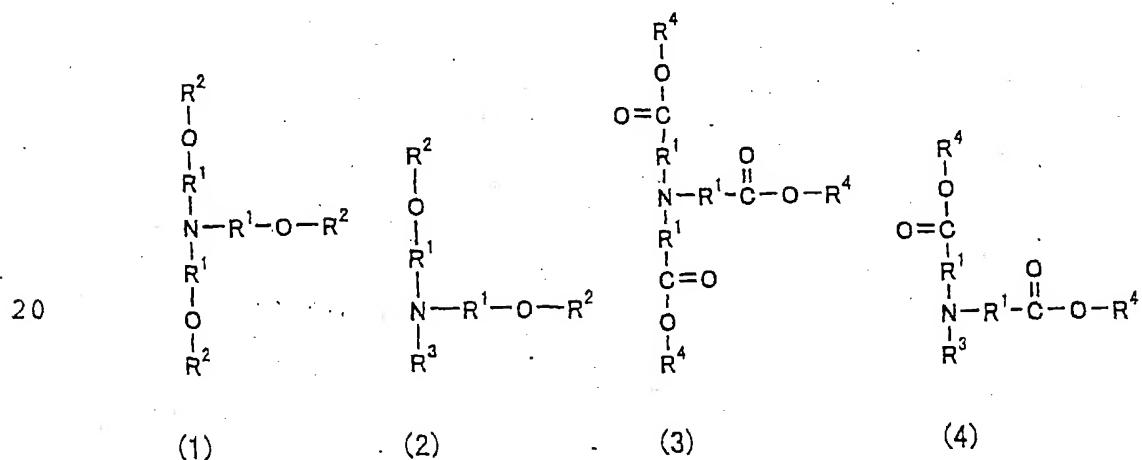


(I)

wherein, n stands for 1, 2 or 3, preferably 1 or 2, side chains Xs may be the same or different and each independently represents  $-R^1-O-R^2$  or  $-R^1-C(=O)-O-R^6$  (in which R's are the

same or different and each independently represents an alkylene group of 1 to 5 carbon atoms, R's are the same or different and each independently represents a linear, branched or cyclic alkyl group of 1 to 20 carbon atoms 5 containing a carbonyl or ester group, and R's are the same or different and each independently represents a linear, branched or cyclic alkyl group of 1 to 20 carbon atoms which may contain a carbonyl group, an ester group, an ether group, a hydroxyl group or a lactone ring, or R<sup>1</sup> and R<sup>2</sup>, or R<sup>1</sup> and R<sup>61</sup> 10 in the same side chain may be coupled together to form a ring); and side chains Ys are the same or different and each independently represents a hydrogen atom or a linear, branched or cyclic alkyl group of 1 to 20 carbon atoms which may contain an ether or hydroxyl group.

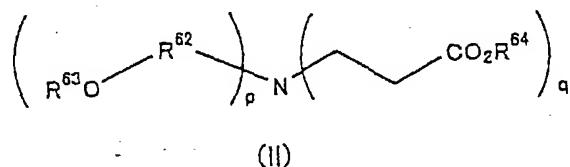
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In the above-described formulas, R's are the same or different and each independently represents an alkylene group 25

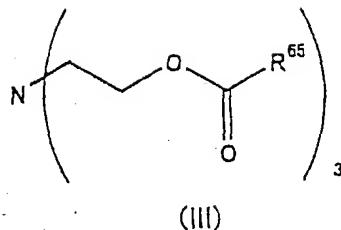
of 1 to 5 carbon atoms, R<sup>2</sup>'s are the same or different and each independently represents a linear, branched or cyclic alkyl group of 1 to 20 carbon atoms containing a carbonyl or ester group, R<sup>3</sup> represents a hydrogen atom or a linear, branched or cyclic alkyl group of 1 to 20 carbon atoms, and R<sup>4</sup>'s are the same or different and each independently represents a linear, branched or cyclic alkyl group of 1 to 20 carbon atoms which may contain a carbonyl, ester or ether group.

10



(II)

wherein, R<sup>62</sup> represents a linear or branched alkylene group of 1 to 5 carbon atoms, p stands for 0, 1 or 2 with the proviso that p+q=3, R<sup>63</sup>'s are the same or different and each independently represents a hydrogen atom or a linear, branched or cyclic alkyl group of 1 to 15 carbon atoms which may contain an ether, carbonyl, ester or hydroxyl group, and R<sup>64</sup>'s are the same or different and each independently represents a linear, branched or cyclic alkyl group of 1 to 20 carbon atoms which may contain a carbonyl group, an ester group, an ether group, a hydroxyl group or a lactone ring.



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wherein,  $\text{R}^{65}$ 's are the same or different and each independently represents a linear, branched or cyclic alkyl groups of 1 to 15 carbon atoms which may contain an ether, carbonyl, ester or hydroxyl group.

10 In another aspect of the present invention, there is also provided a positive type resist composition, which can comprise (A): one or more basic compounds selected from the group consisting of formulas (I) to (III) and formulas (1) to (4), (B): an organic solvent, and (C1): a base resin which is 15 an alkali insoluble or sparingly soluble resin having an acidic functional group protected with an acid-labile group but becomes alkali soluble when the acid-labile group is eliminated, and (D): an acid generator; and optionally (E): a dissolution inhibitor.

20 In a further aspect of the present invention, there is also provided a negative type resist composition, which comprises (A): one or more basic compounds selected from the group consisting of formulas (I) to (III) and (1) to (4), (B): an organic solvent, and (C2): a base resin which is

alkali soluble but becomes alkali sparingly soluble by crosslinking with a crosslinker, (D): an acid generator, and (F): a crosslinker which undergoes crosslinking by an acid.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

5 The present invention relates to a resist composition containing one or more basic compounds selected from those represented by formulas (I) to (III) and (1) to (4).

Specific examples of the basic compounds represented by formulas (I) to (III) and (1) to (4) include, but not limited 10 to, tris(2-formyloxyethyl)amine, tris(2-acetoxyethyl)amine, tris(2-propionyloxyethyl)amine, tris(2-butyryloxyethyl)amine, tris(2-isobutyryloxyethyl)amine, tris(2-valeryloxyethyl)amine, tris(2-pivaloyloxyethyl)amine, N,N-bis(2-acetoxyethyl)-2-(acetoxycetoxy)ethylamine, tris(2-methoxycarbonyloxyethyl)amine, tris(2-tert-butoxycarbonyloxyethyl)amine, tris[2-(2-oxopropoxy)ethyl]amine, tris[2-(methoxycarbonylmethyl)oxyethyl]amine, tris[2-(tert-butoxycarbonylmethyloxy)ethyl]amine, and tris[2-(cyclohexyloxycarbonylmethyloxy)ethyl]amine {corresponding to 15 the formulas (1) and (III)}; tris(2-methoxycarbonylethyl)amine and tris(2-ethoxycarbonylethyl)amine {corresponding to the formula (I), (II) and (3)}; N,N-bis(2-hydroxyethyl)-2-(methoxycarbonyl)ethylamine, N,N-bis(2-acetoxyethyl)-2-

(methoxycarbonyl)ethylamine, N,N-bis(2-hydroxyethyl)-2-  
(ethoxycarbonyl)ethylamine, N,N-bis(2-acetoxyethyl)-2-  
(ethoxycarbonyl)ethylamine, N,N-bis(2-hydroxyethyl)-2-(2-  
methoxyethoxycarbonyl)ethylamine, N,N-bis(2-acetoxyethyl)-2-  
5 (2-methoxyethoxycarbonyl)ethylamine, N,N-bis(2-hydroxyethyl)-  
2-(2-hydroxyethoxycarbonyl)ethylamine, N,N-bis(2-  
acetoxyethyl)-2-(2-acetoxyethoxycarbonyl)ethylamine, N,N-  
bis(2-hydroxyethyl)-2-  
[(methoxycarbonyl)methoxycarbonyl]ethylamine, N,N-bis(2-  
10 acetoxyethyl)-2-[(methoxycarbonyl)methoxycarbonyl]ethylamine,  
N,N-bis(2-hydroxyethyl)-2-(2-oxopropoxycarbonyl)ethylamine,  
N,N-bis(2-acetoxyethyl)-2-(2-oxopropoxycarbonyl)ethylamine,  
N,N-bis(2-hydroxyethyl)-2-  
(tetrahydrofurfuryloxycarbonyl)ethylamine, N,N-bis(2-  
15 acetoxyethyl)-2-(tetrahydrofurfuryloxycarbonyl)ethylamine,  
N,N-bis(2-hydroxyethyl)-2-[(2-oxotetrahydrofuran-3-  
yl)oxycarbonyl]ethylamine, N,N-bis(2-acetoxyethyl)-2-[(2-  
oxotetrahydrofuran-3-yl)oxycarbonyl]ethylamine, N,N-bis(2-  
hydroxyethyl)-2-(4-hydroxybutoxycarbonyl)ethylamine, N,N-  
20 bis(2-formyloxyethyl)-2-(4-  
formyloxybutoxycarbonyl)ethylamine, N,N-bis(2-  
formyloxyethyl)-2-(2-formyloxyethoxycarbonyl)ethylamine, N,N-  
bis(2-methoxyethyl)-2-(methoxycarbonyl)ethylamine, N-(2-  
hydroxyethyl)bis[2-(methoxycarbonyl)ethyl]amine, N-(2-  
25 acetoxyethyl)bis[2-(methoxycarbonyl)ethyl]amine, N-(2-

hydroxyethyl)bis[2-(ethoxycarbonyl)ethyl]amine, N-(2-acetoxyethyl)bis[2-(ethoxycarbonyl)ethyl]amine, N-(3-hydroxy-1-propyl)bis[2-(methoxycarbonyl)ethyl]amine, N-(3-acetoxy-1-propyl)bis[2-(methoxycarbonyl)ethyl]amine, and N-(2-methoxyethyl)bis[2-(methoxycarbonyl)ethyl]amine  
5 {corresponding to the formula (I) and (II)}; N-butylbis[2-(methoxycarbonyl)ethyl]amine and N-butylbis[2-(2-methoxyethoxycarbonyl)ethyl]amine {corresponding to the formula (1) and (4)}; N-methylbis(2-acetoxyethyl)amine, N-10 ethylbis(2-acetoxyethyl)amine, N-methylbis(2-pivaloyloxyethyl)amine, N-ethylbis[2-(methoxycarbonyloxy)ethyl]amine, N-ethylbis[2-(tert-butoxycarbonyloxy)ethyl]amine {corresponding to the formula (I) and (2)}; tris(methoxycarbonylmethyl)amine and  
15 tris(ethoxycarbonylmethyl)amine {corresponding to the formula (I) and (3)}; N-butylbis(methoxycarbonylmethyl)amine and N-hexylbis(methoxycarbonylmethyl)amine {corresponding to the formula (I) and (4)}; and  $\beta$ -(diethylamino)- $\delta$ -valerolactone {corresponding to the formula (I)}.

20 As the organic solvent usable as Component (B) in the present invention, any organic solvent permitting dissolution therein of an acid generator, base resin, dissolution inhibitor or the like may be used. Examples of such an organic solvent include ketones such as cyclohexanone and  
25 methyl-2-n-amylketone, alcohols such as 3-methoxybutanol, 3-

methyl-3-methoxybutanol, 1-methoxy-2-propanol and 1-ethoxy-2-propanol, ethers such as propylene glycol monomethyl ether, ethylene glycol monomethyl ether, propylene glycol monoethyl ether, ethylene glycol monoethyl ether, propylene glycol 5 dimethyl ether and diethylene glycol dimethyl ether and esters such as propylene glycol monomethyl ether acetate, propylene glycol monoethyl ether acetate, ethyl lactate, ethyl pyruvate, butyl acetate, methyl 3-methoxypropionate, ethyl 3-ethoxypropionate, tert-butyl acetate, tert-butyl 10 propionate and propylene glycol mono-tert-butyl ether acetate. They may be used either singly or in admixture of two or more, but an organic solvent usable in the present 15 invention is not limited to them.

In the present invention, diethylene glycol dimethyl ether, 1-ethoxy-2-propanol or ethyl lactate in which the acid generator contained in the resist composition is most soluble, or propylene glycol monomethyl ether acetate which is a safe solvent, or a solvent mixture therewith is 20 preferably employed among the above-described organic solvents.

Regarding a base polymer (C) of the present invention, a base resin (C1) which has an acidic functional group protected with an acid-labile group and is insoluble or sparingly soluble in alkali, but becomes soluble in an alkali 25 by the elimination of the acid-labile group is usable for a

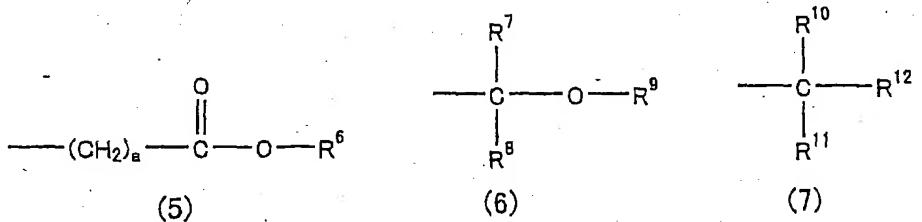
positive type resist composition, while a base resin (C2) which is soluble in alkali but becomes sparingly soluble in alkali by crosslinking with a crosslinker is usable for a negative type resist composition. The "insoluble or sparingly soluble in alkali" means that solubility in an aqueous solution of 2.38% by weight of TMAH (tetramethylammonium hydroxide) is 0 or less than 20 Å/sec, and "soluble in alkali" means that solubility in an aqueous solution of 2.38% by weight of TMAH is 20 to 300 Å/sec.

Examples of the base polymer used as the component (C) in the present invention for a KrF excimer laser resist of 248 nm in wavelength, EUV resist of 3 to 20 nm in wavelength, EB resist or X-ray resist include polyhydroxystyrene (PHS) and copolymers thereof with styrene, (meth)acrylate or maleimide-N-carboxylate; those of the base polymer for an ArF excimer laser of 193 nm in wavelength include (meth)acrylate polymers, alternating copolymers of norbornene and maleic anhydride, polynorbornene polymers and metathesis ring-opening polymerization type polymers; and those for an F<sub>2</sub> excimer laser of 157 nm wavelength include the above-exemplified polymers substituted with fluorine. The base polymer (C) is however not limited thereto. In the above, "(meth)acrylate" means methacrylate or acrylate.

The base polymer (C1) usable for the positive type resist composition is obtained by substituting the hydrogen

atom of the hydroxyl group of the phenol or carboxyl group of the base polymer (C2) with an acid-labile group. It is the common practice to lower the dissolution rate of an unexposed portion by such substitution.

5 As the acid-labile group, groups represented by the below-described formula (5) or (6), tertiary alkyl groups of 4 to 40 carbon atoms represented by the below-described formula (7), trialkylsilyl groups of 1 to 6 carbon atoms and oxoalkyl groups of 4 to 20 carbon atoms are preferred among various 10 groups.



15 wherein,  $\text{R}^6$  and  $\text{R}^9$  each independently represents a linear, branched or cyclic alkyl group of 1 to 20 carbon atoms which may contain a hetero atom such as oxygen, sulfur, nitrogen or fluorine,  $\text{R}^7$  and  $\text{R}^8$  each independently represents a hydrogen atom or a linear, branched or cyclic alkyl group of 1 to 20 carbon atoms which may contain a hetero atom such as oxygen, sulfur, nitrogen or fluorine, and  $\text{a}$  stands for an integer of 0 to 10, or  $\text{R}^7$  and  $\text{R}^8$ ,  $\text{R}^7$  and  $\text{R}^9$ , and  $\text{R}^8$  and  $\text{R}^9$  may each be coupled together to form a ring.

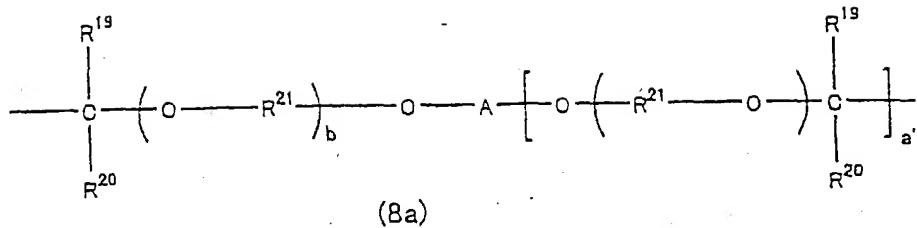
25 Additional examples of  $\text{R}^6$ ,  $\text{R}^9$  or  $\text{R}^{12}$  include non-

substituted or substituted aryl groups such as phenyl group and alkoxy-substituted phenyl groups, e.g., p-methylphenyl, p-ethylphenyl and p-methoxyphenyl and aralkyl groups such as benzyl and phenethyl.

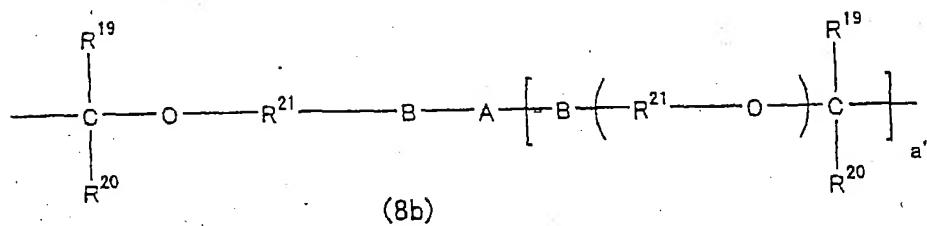
5        Specific examples of the acid-labile group represented by the formula (5) include tert-butoxycarbonyl, tert-butoxycarbonylmethyl, tert-amyoxy carbonyl, tert-amyoxy carbonylmethyl, 1-ethoxyethoxy carbonylmethyl, 2-tetrahydropyranoyloxy carbonylmethyl and 2-tetrahydrofuranoyloxy carbonylmethyl groups.

10      Specific examples of the acid-labile group represented by the formula (6) include linear or branched acetal groups such as 1-methoxyethyl, 1-ethoxyethyl, 1-n-propoxyethyl, 1-isopropoxyethyl, 1-n-butoxyethyl, 1-isobutoxyethyl, 1-sec-butoxyethyl, 1-tert-butoxyethyl, 1-tert-amyoxyethyl, 1-ethoxy-n-propyl, 1-cyclohexyloxyethyl, methoxypropyl, ethoxypropyl, 1-methoxy-1-methylethyl and 1-ethoxy-1-methylethyl; and cyclic acetal groups such as tetrahydrofuryl and tetrahydropyranyl groups. Preferred are ethoxyethyl, butoxyethyl and ethoxypropyl groups.

15      At least 0.1% of the hydrogen atoms of the hydroxyl groups of the base polymer may undergo intramolecular or intermolecular crosslinking via an acid-labile group (crosslinking group) represented by the following formula (8a) or (8b).



5



(8b)

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In the above formulas,  $\text{R}^{19}$  and  $\text{R}^{20}$  each represents a hydrogen atom or a linear, branched or cyclic alkyl group of 1 to 8 carbon atoms.  $\text{R}^{19}$  and  $\text{R}^{20}$  may be coupled together to form a ring and if so,  $\text{R}^{19}$  and  $\text{R}^{20}$  each represents a linear or 15 branched  $\text{C}_{1-8}$  alkylene group.

$\text{R}^{21}$  represents a linear, branched or cyclic alkylene group of 1 to 10 carbon atoms and  $b$  stands for 0 or an integer of 1 to 10.

A represents an aliphatic or alicyclic, saturated hydrocarbon group, aromatic hydrocarbon group or heterocyclic group having a valence of  $a$  and 1 to 50 carbon atoms. The above-described group may have a hetero atom interposed therein, or a part of the hydrogen atoms bonded to the carbon atoms of the group may be substituted with a hydroxyl group, 25 a carboxyl group, a carbonyl group or a fluorine atom. B

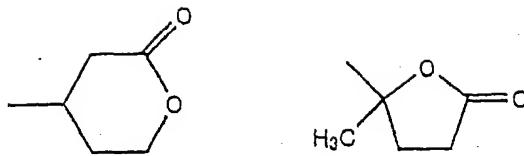
represents  $-\text{CO}-\text{O}-$ ,  $-\text{NHCO}-\text{O}-$  or  $\text{NHCONH}-$ .  $a$  stands for an integer of 2 to 8, and  $a'$  represents an integer of 1 to 7. They satisfy the following equation of  $a = (a' + 1)$ .

$R^{10}$ ,  $R^{11}$  and  $R^{12}$  in the formula (7) each represents a 5 monovalent hydrocarbon group such as linear, branched or cyclic  $C_{1-20}$  alkyl group and it may contain a hetero atom such as oxygen, sulfur, nitrogen or fluorine.  $R^{10}$  and  $R^{11}$ ,  $R^{10}$  and  $R^{12}$ , or  $R^{11}$  and  $R^{12}$  may be coupled together to form a ring. Alternatively, these groups may contain an oxygen atom, their 10 hydrogen atom bonded to the carbon atom may be substituted with a hydroxyl group, or their two hydrogen atoms may be substituted with an oxygen atom to form a carbonyl-group.

Examples of the tertiary alkyl group represented by the formula (7) include tert-butyl, triethylcarbinyl, 2-ethyl-2-15 norbornyl, 1-methylcyclohexyl, 1-ethylcyclopentyl, 2-(2-methyl)adamantyl, 2-(2-ethyl)adamantyl and tert-amyl groups.

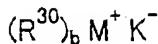
Examples of the trialkylsilyl group of 1 to 6 carbon atoms used as the acid-labile group include trimethylsilyl, triethylsilyl and dimethyl-tert-butyldisilyl groups.

20 Examples of the oxoalkyl group of 4 to 20 carbon atoms used as the acid-labile group include a 3-oxocyclohexyl group and groups represented by the following formulas:

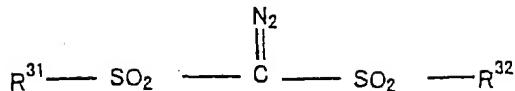


5 The base polymer as Component (C) preferably has a weight average molecular weight ranging from 5,000 to 100,000. The molecular weights less than 5,000 may make the film forming property and resolution property inferior, while the molecular weights exceeding 100,000, on the other hand, 10 may make resolution property inferior.

Examples of the acid generator as Component (D) include onium salts of the below-described formula (9), diazomethane derivatives of the formula (10), glyoxime derivatives of the formula (11),  $\beta$ -ketosulfone derivatives, disulfone derivatives, nitrobenzyl sulfonate derivatives, sulfonate derivatives and imido-ylsulfonate derivatives. 15

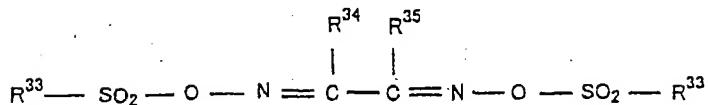


(9)



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(10)



- (11)

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In the above-described formulas,  $R^{30}$ 's may be the same or different and each independently represents a linear, branched or cyclic alkyl group of 1 to 12 carbon atoms, aryl group of 6 to 12 carbon atoms or aralkyl group of 7 to 12 carbon atoms,  $M^+$  represents iodonium or sulfonium,  $K^-$  represents a non-nucleophilic counterion and  $b$  stands for 2 or 3.

$R^{31}$  and  $R^{32}$  each represents a linear, branched or cyclic alkyl group of 1 to 12 carbon atoms or halogenated alkyl group, aryl or halogenated aryl group of 6 to 12 carbon atoms, or aralkyl group of 7 to 12 carbon atoms.

$R^{33}$ ,  $R^{34}$  and  $R^{35}$  each represents a linear, branched or cyclic alkyl group or halogenated alkyl group of 1 to 12 carbon atoms, aryl or halogenated aryl group of 6 to 12 carbon atoms, or an aralkyl group of 7 to 12 carbon atoms.

$R^{34}$  and  $R^{35}$  may be coupled together to form a cyclic structure and if so,  $R^{34}$  and  $R^{35}$  each represents a linear or branched alkylene group of 1 to 6 carbon atoms.

Examples of the alkyl group of 1 to 12 carbon atoms represented by  $R^{30}$  in the formula (9) include methyl, ethyl, propyl, butyl, cyclohexyl, 2-oxocyclohexyl, norbornyl and adamantyl groups. Examples of the aryl group of 6 to 12 carbon atoms include phenyl, alkoxyphenyl groups such as p-methoxyphenyl, m-methoxyphenyl, o-methoxyphenyl, ethoxyphenyl, p-tert-butoxyphenyl and m-tert-butoxyphenyl and alkylphenyl groups such as 2-methylphenyl, 3-methylphenyl, 4-methylphenyl, ethylphenyl, 4-tert-butylphenyl, 4-butylphenyl and dimethylphenyl groups. Examples of the aralkyl groups of 7 to 12 carbon atoms include benzyl and phenetyl. Examples of the non-nucleophilic counterion as  $K^-$  include halide ions such as chloride ion and bromide ion, fluoroalkyl sulfonates such as triflate, 2,2,2-trifluoroethane sulfonate and nonafluorobutane sulfonate, aryl sulfonates such as tosylate, benzene sulfonate, 4-fluorobenzene sulfonate and 2,3,4,5,6-pentafluorobenzene sulfonate and alkyl sulfonates such as mesylate and butane sulfonate.

Examples of the alkyl group of 1 to 12 carbon atoms as  $R^{31}$  or  $R^{32}$  in the formula (10) include methyl, ethyl, propyl, butyl, amyl, cyclopentyl, cyclohexyl, norbornyl and adamantyl groups; those of the halogenated alkyl group of 1 to 12

carbon atoms include trifluoromethyl, 2,2,2-trifluoroethyl, 2,2,2-trichloroethyl and nonafluorobutyl groups; those of the C<sub>6-12</sub> aryl group include a phenyl group and alkoxyphenyl groups such as p-methoxyphenyl, m-methoxyphenyl, o-methoxyphenyl, ethoxyphenyl, p-tert-butoxyphenyl and m-tert-butoxyphenyl groups and alkylphenyl groups such as 2-methylphenyl, 3-methylphenyl, 4-methylphenyl, ethylphenyl, 4-tert-butylphenyl, 4-butylphenyl and dimethylphenyl groups; those of the halogenated aryl groups of 6 to 12 carbon atoms include fluorobenzene, chlorobenzene and 2,3,4,5,6-pentafluorobenzene groups; and those of the aralkyl group of 7 to 12 carbon atoms include benzyl and phenetyl.

Examples of the alkyl group of 1 to 12 carbon atoms, halogenated alkyl group of 1 to 12 carbon atoms, aryl group of 6 to 12 carbon atoms, halogenated aryl group of 6 to 12 carbon atoms and aralkyl group of 7 to 12 carbon atoms represented by R<sup>33</sup>, R<sup>34</sup> or R<sup>35</sup> in the formula (11), include those described in R<sup>31</sup> and R<sup>32</sup>. Examples of the alkylene group of 1 to 6 carbon atoms for R<sup>34</sup> or R<sup>35</sup> include methylene, ethylene, propylene, butylene and hexylene groups.

Specific examples of the onium salt in the formula (9) include diphenyliodonium trifluoromethanesulfonate, (p-tert-butoxyphenyl)phenyliodonium trifluoromethanesulfonate, diphenyliodonium p-toluenesulfonate, (p-tert-butoxyphenyl)phenyliodonium p-toluenesulfonate,

triphenylsulfonium trifluoromethanesulfonate, (p-tert-  
butoxyphenyl)diphenylsulfonium trifluoromethanesulfonate,  
bis(p-tert-butoxyphenyl)phenylsulfonium  
trifluoromethanesulfonate, tris(p-tert-butoxyphenyl)sulfonium  
5 trifluoromethanesulfonate, triphenylsulfonium p-  
toluenesulfonate, (p-tert-butoxyphenyl)diphenylsulfonium p-  
toluenesulfonate, bis(p-tert-butoxyphenyl)phenylsulfonium p-  
toluenesulfonate, tris(p-tert-butoxyphenyl)sulfonium p-  
toluenesulfonate, triphenylsulfonium  
10 nonafluorobutanesulfonate, triphenylsulfonium  
butanesulfonate, trimethylsulfonium  
trifluoromethanesulfonate, trimethylsulfonium p-  
toluenesulfonate, cyclohexylmethyl(2-oxocyclohexyl)sulfonium  
trifluoromethanesulfonate, cyclohexylmethyl(2-  
15 oxocyclohexyl)sulfonium p-toluenesulfonate,  
dimethylphenylsulfonium trifluoromethanesulfonate,  
dimethylphenylsulfonium p-toluenesulfonate,  
dicyclohexylphenylsulfonium trifluoromethanesulfonate and  
dicyclohexylphenylsulfonium p-toluenesulfonate.  
20 Examples of the diazomethane derivative in the formula  
(10) include bis(benzenesulfonyl)diazomethane, bis(p-  
toluenesulfonyl)diazomethane,  
bis(xylenesulfonyl)diazomethane,  
bis(cyclohexylsulfonyl)diazomethane,  
25 bis(cyclopentylsulfonyl)diazomethane, bis(n-

butylsulfonyl)diazomethane,  
bis(isobutylsulfonyl)diazomethane, bis(sec-  
butylsulfonyl)diazomethane, bis(n-  
propylsulfonyl)diazomethane,  
5 bis(isopropylsulfonyl)diazomethane, bis(tert-  
butylsulfonyl)diazomethane, bis(n-amylsulfonyl)diazomethane,  
bis(isoamylsulfonyl)diazomethane, bis(sec-  
amylsulfonyl)diazomethane, bis(tert-  
amylsulfonyl)diazomethane, 1-cyclohexylsulfonyl-1-(tert-  
10 butylsulfonyl)diazomethane, 1-cyclohexylsulfonyl-1-(tert-  
amylsulfonyl)diazomethane and 1-tert-amylsulfonyl-1-(tert-  
butylsulfonyl)diazomethane.

Examples of the glyoxime derivative in the formula (11) include bis-o-(p-toluenesulfonyl)- $\alpha$ -dimethylglyoxime, bis-o-(p-toluenesulfonyl)- $\alpha$ -diphenylglyoxime, bis-o-(p-toluenesulfonyl)- $\alpha$ -dicyclohexylglyoxime, bis-o-(p-toluenesulfonyl)-2,3-pentanedioneglyoxime, bis-o-(p-toluenesulfonyl)-2-methyl-3,4-pentanedioneglyoxime, bis-o-(n-butanesulfonyl)- $\alpha$ -dimethylglyoxime, bis-o-(n-butanesulfonyl)- $\alpha$ -diphenylglyoxime, bis-o-(n-butanesulfonyl)- $\alpha$ -dicyclohexylglyoxime, bis-o-(n-butanesulfonyl)-2,3-pentanedioneglyoxime, bis-o-(n-butanesulfonyl)-2-methyl-3,4-pentanedioneglyoxime, bis-o-(methanesulfonyl)- $\alpha$ -dimethylglyoxime, bis-o-(trifluoromethanesulfonyl)- $\alpha$ -dimethylglyoxime, bis-o-(1,1,1-

trifluoroethanesulfonyl)- $\alpha$ -dimethylglyoxime, bis- $\alpha$ -(tert-  
butanesulfonyl)- $\alpha$ -dimethylglyoxime, bis- $\alpha$ -  
(perfluorooctanesulfonyl)- $\alpha$ -dimethylglyoxime, bis- $\alpha$ -  
(cyclohexanesulfonyl)- $\alpha$ -dimethylglyoxime, bis- $\alpha$ -  
5 (benzenesulfonyl)- $\alpha$ -dimethylglyoxime, bis- $\alpha$ -(p-  
fluorobenzenesulfonyl)- $\alpha$ -dimethylglyoxime, bis- $\alpha$ -(p-tert-  
butylbenzenesulfonyl)- $\alpha$ -dimethylglyoxime, bis- $\alpha$ -  
(xylenesulfonyl)- $\alpha$ -dimethylglyoxime and bis- $\alpha$ -  
(camphorsulfonyl)- $\alpha$ -dimethylglyoxime.

10 Examples of the  $\beta$ -ketosulfone derivative include 2-  
cyclohexylcarbonyl-2-(p-toluenesulfonyl)propane and 2-  
isopropylcarbonyl-2-(p-toluenesulfonyl)propane.

Further examples of the acid generator (D) include  
disulfone derivatives such as diphenyldisulfone and  
15 dicyclohexyldisulfone, nitrobenzyl sulfonate derivatives such  
as 2,6-dinitrobenzyl p-toluenesulfonate and 2,4-dinitrobenzyl  
p-toluenesulfonate, sulfonate derivatives such as 1,2,3-  
tris(methanesulfonyloxy)benzene, 1,2,3-  
tris(trifluoromethanesulfonyloxy)benzene and 1,2,3-tris(p-  
20 toluenesulfonyloxy)benzene, imido-ylsulfonate derivatives  
such as phthalimido-yl-triflate, phthalimido-yl-tosylate, 5-  
norbornene-2,3-dicarboxyimido-yl-triflate, 5-norbornene-2,3-  
dicarboxyimido-yl-tosylate and 5-norbornene-2,3-  
dicarboxyimido-yl-n-butylsulfonate.

25 Preferred examples include onium salts such as

triphenylsulfonium trifluoromethanesulfonate, (p-tert-  
butoxyphenyl)diphenylsulfonium trifluoromethanesulfonate,  
tris(p-tert-butoxyphenyl)sulfonium trifluoromethanesulfonate,  
triphenylsulfonium p-toluenesulfonate, (p-tert-  
5 butoxyphenyl)diphenylsulfonium p-toluenesulfonate, and  
tris(p-tert-butoxyphenyl)sulfonium p-toluenesulfonate;  
diazomethane derivatives such as  
bis(benzenesulfonyl)diazomethane, bis(p-  
toluenesulfonyl)diazomethane,  
10 bis(cyclohexylsulfonyl)diazomethane, bis(n-  
butylsulfonyl)diazomethane,  
bis(isobutylsulfonyl)diazomethane, bis(sec-  
butylsulfonyl)diazomethane, bis(n-  
propylsulfonyl)diazomethane,  
15 bis(isopropylsulfonyl)diazomethane, and bis(tert-  
butylsulfonyl)diazomethane; glyoxime derivatives such as bis-  
o-(p-toluenesulfonyl)- $\alpha$ -dimethylglyoxime and bis-o-(n-  
butanesulfonyl)- $\alpha$ -dimethylglyoxime; and  
naphthoquinonediazidesulfonate derivatives.

20 The above-exemplified acid generators may be used either  
singly or in combination.

The onium salts are excellent in rectangularity  
improving effects, while the diazomethane derivatives and  
glyoxime derivatives are excellent in standing-wave reducing  
25 effects. Use of these two types of acid generators in

combination makes it possible to conduct fine profile control.

The acid generator is preferably added in an amount of 0.2 to 50 parts by weight, particularly, 0.5 to 40 parts by weight based on 100 parts by weight of the base resin in total. When the amount is less than 0.2 part by weight, an acid generation amount upon exposure is small, which may lead to inferior sensitivity and resolution. Amounts exceeding 50 parts by weight, on the other hand, lower the transmittance of the resulting resist, which may lead to inferior resolution.

Examples of the dissolution inhibitor serving as the component (E) include compounds having a molecular weight of 3000 or less whose solubility in an alkali developing solution changes by acid action, particularly, compounds obtained by partially or wholly substituting a phenol or carboxylic acid derivative having a molecular weight as low as 2,500 or less with an acid-labile substituent.

Examples of the phenol or carboxylic acid derivative having a molecular weight of 2500 or less include bisphenol A, bisphenol H, bisphenol S, 4,4-bis(4'-hydroxyphenyl)valeric acid, tris(4-hydroxyphenyl)methane, 1,1,1-tris(4'-hydroxyphenyl)ethane, 1,1,2-tris(4'-hydroxyphenyl)ethane, phenolphthalein, and thymolphthalein, while those of the acid-labile substituent include those exemplified above as

the acid-labile group of the base polymer.

Preferred examples of the dissolution inhibitor include  
bis(4-(2'-tetrahydropyranyloxy)phenyl)methane, bis(4-(2'-  
tetrahydrofuranyloxy)phenyl)methane, bis(4-tert-  
5 butoxyphenyl)methane, bis(4-tert-  
butoxycarbonyloxyphenyl)methane, bis(4-tert-  
butoxycarbonylmethyloxyphenyl)methane, bis(4-(1'-  
ethoxyethoxy)phenyl)methane, bis(4-(1'-  
ethoxypropyloxy)phenyl)methane, 2,2-bis(4'-(2"-  
10 tetrahydropyranyloxy))propane, 2,2-bis(4'-(2"-  
tetrahydrofuranyloxy)phenyl)propane, 2,2-bis(4'-tert-  
butoxyphenyl)propane, 2,2-bis(4'-tert-  
butoxycarbonyloxyphenyl)propane, 2,2-bis(4'-tert-  
butoxycarbonylmethyloxyphenyl)propane, 2,2-bis(4'-(1"-  
15 ethoxyethoxy)phenyl)propane, 2,2-bis(4'-(1"-  
ethoxypropyloxy)phenyl)propane, tert-butyl 4,4-bis(4'-(2"-  
tetrahydropyranyloxy)phenyl)valerate, tert-butyl 4,4-bis(4'-(2"-  
20 tetrahydrofuranyloxy)phenyl)valerate, tert-butyl 4,4-bis(4'-tert-  
butoxycarbonyloxyphenyl)valerate, tert-butyl 4,4-bis(4'-tert-  
butoxycarbonylmethyloxyphenyl)valerate, tert-butyl 4,4-  
bis(4'-(1"-ethoxyethoxy)phenyl)valerate, tert-butyl 4,4-  
bis(4'-(1"-ethoxypropyloxy)phenyl)valerate, tris(4-(2'-  
25 tetrahydropyranyloxy)phenyl)methane, tris(4-(2'-  
tetrahydrofuranyloxy)phenyl)methane, tris(4-tert-

butoxyphenyl)methane, tris(4-tert-  
butoxycarbonyloxyphenyl)methane, tris(4-tert-  
butoxycarbonyloxymethylphenyl)methane, tris(4-(1'-  
ethoxyethoxy)phenyl)methane, tris(4-(1'-  
5 ethoxypropyloxy)phenyl)methane, 1,1,2-tris(4'-(2'-  
tetrahydropyranyloxy)phenyl)ethane, 1,1,2-tris(4'-(2'-  
tetrahydrofuryloxy)phenyl)ethane, 1,1,2-tris(4'-tert-  
butoxyphenyl)ethane, 1,1,2-tris(4'-tert-  
butoxycarbonyloxyphenyl)ethane, 1,1,2-tris(4'-tert-  
10 butoxycarbonylmethyloxyphenyl)ethane, 1,1,2-tris(4'-(1'-  
ethoxyethoxy)phenyl)ethane, and 1,1,2-tris(4'-(1'-  
ethoxypropyloxy)phenyl)ethane.

In the present invention, the dissolution inhibitor is added preferably in an amount of 20 parts by weight or less, 15 more preferably 15 parts by weight or less relative to 100 parts by weight of the solid content in the resist material. Amounts greater than 20 parts by weight increase the monomer component, may thereby deteriorating the heat resistance of the resist material.

20 Examples of the crosslinker as Component (F) include compounds having, in the molecule thereof, at least two hydroxymethyl groups, alkoxymethyl groups, epoxy groups or vinyl ether groups. Among them, substituted glycoluril derivatives, urea derivatives and hexa(methoxymethyl)melamine 25 are used preferably.

Specific examples include N,N,N',N'-tetramethoxymethylurea, hexamethylmelamine, tetrahydroxymethyl-substituted glycolurils, tetraalkoxymethyl-substituted glycolurils such as 5 tetramethoxymethyl glycoluril, bis-hydroxymethylphenols, and condensation products between phenol compounds and epichlorohydrin or the like such as bisphenol A.

Particularly preferred examples of the crosslinker include 1,3,5,7-tetraalkoxymethylglycolurils such as 1,3,5,7-10 tetramethoxymethylglycoluril, 1,3,5,7-tetrahydroxymethylglycoluril, 2,6-dihydroxymethyl p-cresol, 2,6-dihydroxymethylphenol, 2,2',6,6'-tetrahydroxymethyl bisphenol A, 1,4-bis-[2-(2-hydroxypropyl)]-benzene, N,N,N',N'-tetramethoxymethylurea and 15 hexamethoxymethylmelamine.

A crosslinker which undergoes crosslinking by an acid is preferred.

The amount of the crosslinker is not limited, but it is added in an amount of 1 to 25 parts by weight, preferably 5 to 20 parts by weight based on the whole solid content of the resist composition. The above-described crosslinkers may be used either singly or in combination.

As a base, conventionally employed bases other than the above-described ones may be used either singly or in 25 combination. Examples of the conventionally employed bases

include aliphatic primary, secondary or tertiary amines, mixed amines, aromatic amines, heterocyclic amines, nitrogen-containing compounds having a carboxyl group, nitrogen-containing compounds having a sulfonyl group, nitrogen-containing compounds having a hydroxy group, nitrogen-containing compounds having a hydroxyphenyl group, alcoholic nitrogen-containing compounds, amide derivatives and imide derivatives.

5 Examples of the aliphatic primary amine include ammonia, methylamine, ethylamine, n-propylamine, isopropylamine, n-butylamine, isobutylamine, sec-butylamine, tert-butylamine, pentylamine, tert-amylamine, cyclopentylamine, hexylamine, cyclohexylamine, heptylamine, octylamine, nonylamine, decylamine, dodecylamine, cetylamine, methylenediamine, 10 ethylenediamine, and tetraethylenepentamine, those of the aliphatic secondary amine include dimethylamine, diethylamine, di-n-propylamine, diisopropylamine, di-n-butylamine, diisobutylamine, di-sec-butylamine, dipentylamine, dicyclopentylamine, dihexylamine, 15 dicyclohexylamine, diheptylamine, dioctylamine, dinonylamine, didecylamine, didodecylamine, dicetylamine, N,N-dimethylmethylenediamine, N,N-dimethylethylenediamine and N,N-dimethyltetraethylenepentamine; and those of the aliphatic tertiary amine include trimethylamine, 20 triethylamine, tri-n-propylamine, triisopropylamine, tri-n-

butylamine, triisobutylamine, tri-sec-butylamine,  
tripentylamine, tricyclopentylamine, trihexylamine,  
tricyclohexylamine, triheptylamine, trioctylamine,  
trinonylamine, tridecylamine, tridodecylamine, tricetylamine,  
5 N,N,N',N'-tetramethylmethylenediamine, N,N,N',N'-  
tetramethylenediamine and N,N,N',N'-  
tetramethyltetraethylenepentamine.

Examples of the mixed amine include dimethylethylamine,  
methylmethylethylpropylamine, benzylamine, phenethylamine and  
10 benzyldimethylamine. Specific examples of the aromatic amine  
or heterocyclic amine include aniline derivatives (ex.  
aniline, N-methylaniline, N-ethylaniline, N-propylaniline,  
N,N-dimethylaniline, 2-methylaniline, 3-methylaniline, 4-  
methylaniline, ethylaniline, propylaniline, trimethylaniline,  
15 2-nitroaniline, 3-nitroaniline, 4-nitroaniline, 2,4-  
dinitroaniline, 2,6-dinitroaniline, 3,5-dinitroaniline, N,N-  
dimethyltoluidine, etc.), diphenyl(p-tolyl)amine,  
methyldiphenylamine, triphenylamine, phenylenediamine,  
naphthylamine, diaminonaphthalene, pyrrole derivatives (ex.  
20 pyrrole, 2H-pyrrole, 1-methylpyrrole, 2,4-dimethylpyrrole,  
2,5-dimethylpyrrole, N-methylpyrrole, etc.), oxazole  
derivatives (ex. oxazole, isoxazole, etc.), thiazole  
derivatives (ex. thiazole, isothiazole, etc.), imidazole  
derivatives (ex. imidazole, 4-methylimidazole, 4-methyl-2-  
25 phenylimidazole, etc.), pyrazole derivatives, furazane

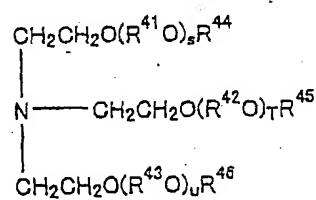
derivatives, pyrroline derivatives (ex. pyrroline, 2-methyl-1-pyrroline, etc.), pyrrolidine derivatives (ex. pyrrolidine, N-methylpyrrolidine, pyrrolidinone, N-methylpyrrolidone, etc.), imidazoline derivatives, imidazolidine derivatives,  
5 pyridine derivatives (ex. pyridine, methylpyridine, ethylpyridine, propylpyridine, butylpyridine, 4-(1-butylpentyl)pyridine, dimethylpyridine, trimethylpyridine, triethylpyridine, phenylpyridine, 3-methyl-2-phenylpyridine, 4-tert-butylpyridine, diphenylpyridine, benzylpyridine,  
10 methoxypyridine, butoxypyridine, dimethoxypyridine, 1-methyl-2-pyridone, 4-pyrrolidinopyridine, 1-methyl-4-phenylpyridine, 2-(1-ethylpropyl)pyridine, amonopyridine, dimethylaminopyridine, etc.), pyridazine derivatives, pyrimidine derivatives, pyrazine derivatives, pyrazoline  
15 derivatives, pyrazolidine derivatives, piperidine derivatives, piperazine derivatives, morpholine derivatives, indole derivatives, isoindole derivatives, 1H-indazole derivatives, indoline derivatives, quinoline derivatives (ex. quinoline, 3-quinolincarbonitrile, etc.), isoquinoline derivatives, cinnoline derivatives, quinazoline derivatives, quinoxaline derivatives, phthalazine derivatives, purine derivatives, pteridine derivatives, carbazole derivatives, phenanthridine derivatives, acridine derivatives, phenazine derivatives, 1,10-phenanthroline derivatives, adenine  
20 derivatives, adenosine derivatives, guanine derivatives,  
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guanosine derivatives, uracil derivatives and uridine derivatives.

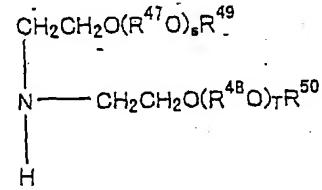
Examples of the nitrogen-containing compound having a carboxyl group include aminobenzoic acid, indolecarboxylic acid, amino acid derivatives (ex. nicotinic acid, alanine, arginine, asparatic acid, glutamic acid, glycine, histidine, isoleucine, glycylleucine, leucine, methionine, phenylalanine, threonine, lysine, 3-aminopyrazine-2-carboxylic acid, methoxyalanine, etc.); those of the nitrogen containing compound having a sulfonyl group include 3-pyridinesulfonic acid and pyridinium p-toluenesulfonate; and those of the nitrogen-containing compound having a hydroxy group, nitrogen-containing compound having a hydroxyphenyl group or alcoholic nitrogen-containing compound include 2-hydroxypyridine, aminocresol, 2,4-quinolinediol, 3-indolemethanol hydrate, monoethanolamine, diethanolamine, triethanolamine, N-ethyldiethanolamine, N,N-diethylethanolamine, triisopropanolamine, 2,2'-iminodiethanol, 2-aminoethanol, 3-amino-1-propanol, 4-amino-1-butanol, 4-(2-hydroxyethyl)morpholine, 2-(2-hydroxyethyl)pyridine, 1-(2-hydroxyethyl)piperazine, 1-[2-(2-hydroxyethoxy)ethyl]piperazine, piperidineethanol, 1-(2-hydroxyethyl)pyrrolidine, 1-(2-hydroxyethyl)-2-pyrrolidinone, 3-piperidino-1,2-propanediol, 3-pyrrolidino-1,2-propanediol, 8-hydroxyjulolidine, 3-quinuclidinol, 3-tropanol, 1-methyl-2-

pyrrolidineethanol, 1-aziridineethanol, N-(2-hydroxyethyl)phthalimide and N-(2-hydroxyethyl)isonicotinamide. Examples of the amide derivative include formamide, N-methylformamide, N,N-dimethylformamide, acetamide, N-methylacetamide, N,N-dimethylacetamide, propionamide, and benzamide, while those of the imide derivatives include phthalimide, succinimide and maleimide.

Furthermore, the basic compound represented by the  
10 below-described formula (B)-1 or (B)-2 can be incorporated.



15 (B)-1



(B)-2

wherein,  $\text{R}^{41}$ ,  $\text{R}^{42}$ ,  $\text{R}^{43}$ ,  $\text{R}^{47}$  and  $\text{R}^{48}$  each independently represents a linear, branched or cyclic  $\text{C}_{1-20}$  alkylene group,  $\text{R}^{44}$ ,  $\text{R}^{45}$ ,  $\text{R}^{46}$ ,  $\text{R}^{49}$  and  $\text{R}^{50}$  each independently represents a hydrogen atom, a  $\text{C}_{1-20}$  alkyl group or an amino group, or  $\text{R}^{44}$  and  $\text{R}^{45}$ ,  $\text{R}^{45}$  and  $\text{R}^{46}$ ,  $\text{R}^{44}$  and  $\text{R}^{46}$ ,  $\text{R}^{44}$  with  $\text{R}^{45}$  and  $\text{R}^{46}$ , and  $\text{R}^{49}$  and  $\text{R}^{50}$  may each be coupled together to form a ring, and  $S$ ,  $T$  and  $U$  each independently stands for an integer of 0 to 20 with the proviso that when  $S$ ,  $T$ ,  $U=0$ , each of  $\text{R}^{44}$ ,  $\text{R}^{45}$ ,  $\text{R}^{46}$ ,  $\text{R}^{49}$  and  $\text{R}^{50}$  does not contain a hydrogen atom.

The alkylene group for R<sup>41</sup>, R<sup>42</sup>, R<sup>43</sup>, R<sup>47</sup> or R<sup>48</sup> contains 1 to 20 carbon atoms, preferably 1 to 10 carbon atoms, more preferably 1 to 8 carbon atoms. Specific examples include 5 methylene, ethylene, n-propylene, isopropylene, n-butylene, isobutylene, n-pentylene, isopentylene, hexylene, nonylene, decylene, cyclopentylene and cyclohexylene groups.

The alkyl group for R<sup>44</sup>, R<sup>45</sup>, R<sup>46</sup>, R<sup>49</sup> or R<sup>50</sup>, contains 1 to 20 carbon atoms, preferably 1 to 8 carbon atoms, more preferably 1 to 6 carbon atoms. Specific examples include 10 methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, n-pentyl, isopentyl, hexyl, nonyl, decyl, dodecyl, tridecyl, cyclopentyl and cyclohexyl groups.

When R<sup>44</sup> and R<sup>45</sup>, R<sup>45</sup> and R<sup>46</sup>, R<sup>44</sup> and R<sup>46</sup>, R<sup>44</sup> and R<sup>45</sup> and 15 R<sup>46</sup>, or R<sup>49</sup> and R<sup>50</sup> form a ring, the ring has 1 to 20, preferably 1 to 8, more preferably 1 to 6 carbon atoms. In this ring, a C<sub>1-6</sub>, particularly C<sub>1-4</sub> alkyl group may be branched.

S, T and U each stands for an integer of 0 to 20, preferably 1 to 10, more preferably 1 to 8.

20 Specific examples of the Compound (B)-1 or (B)-2 include tris{2-(methoxymethoxy)ethyl}amine, tris{2-(methoxyethoxy)ethyl}amine, tris[2-[(2-methoxyethoxy)methoxy]ethyl]amine, tris{2-(2-methoxyethoxy)ethyl}amine, tris{2-(1-methoxyethoxy)ethyl}amine, tris{2-(1-

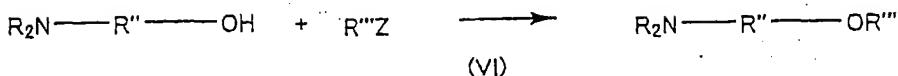
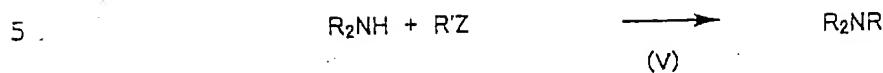
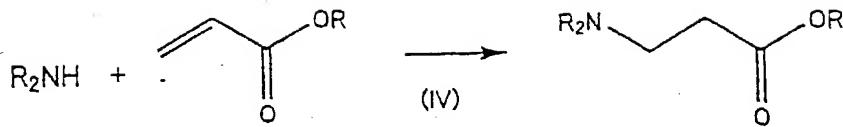
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ethoxyethoxy)ethyl}amine, tris{2-(1-ethoxypropoxy)ethyl}amine, tris[2-((2-hydroxyethoxy)ethoxy)ethyl]amine, 4,7,13,16,21,24-hexaoxa-1,10-diazabicyclo[8.8.8]hexacosane, 4,7,13,18-tetraoxa-1,10-diazabicyclo[8.5.5]eicosane, 1,4,10,13-tetraoxa-7,16-diazabicyclooctadecane, and 1-aza-12-crown-4,1-aza-15-crown-5,1-aza-18-crown-6.

In the present invention, the basic compound is preferably incorporated in an amount of 0.001 to 2 parts by weight, particularly 0.01 to 1 part by weight relative to 100 parts by weight of the whole basic resin. Amounts less than 0.001 part by weight may not bring about any incorporating effects, while those exceeding 2 parts by weight may excessively lower the sensitivity.

For efficient synthesis of the base of the present invention, Michael addition (exemplified by the formula IV) or alkylation (exemplified by the formula V) of an amine, and acylation or etherification (exemplified by the formula VI) of an alkanolamine are used either singly or in combination.

The synthesis process is however not limited thereto.



10

wherein, R and R' may be the same or different and each independently represents a substituted or unsubstituted methyl group, R'' represents an alkylene group, R''' represents a substituted or unsubstituted methyl group (in the case of etherification) or a substituted or unsubstituted acyl group (in the case of acylation), and Z represents a leaving group such as halogen.

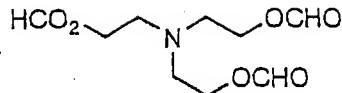
#### Examples

The present invention will hereinafter be described more specifically by Synthesis Examples, Examples and Comparative Examples. It should however be borne in mind that the present invention is not limited to or by them.

The basic compound of the present invention was synthesized in the below-described manner.

25 [Synthesis Example 1] Synthesis of tris (2-

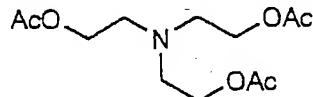
formyloxyethyl)amine:



A mixture of 6.00 g of triethanolamine and 200 g of 5 formic acid was heated to 100°C for 20 hours. After removal of the formic acid by distillation under reduced pressure, the residue was diluted with ethyl acetate, followed by neutralization with a saturated aqueous solution of sodium bicarbonate. After separation into layers, washing with 10 water and drying over anhydrous sodium sulfate, concentration was conducted under reduced pressure. The residue was purified by distillation under reduced pressure, whereby 8.10 g of tris(2-formyloxyethyl)amine was obtained (boiling point: 127 to 131°C/60 Pa, yield: 86%).

15 IR (thin film):  $\nu$  = 2954, 2839, 1722, 1458, 1362, 1282, 1165, 1076, 1005  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta$  = 2.86 (6H, t,  $J$ =5.9Hz), 4.19 (6H, t,  $J$ =5.9Hz), 8.03 (3H, s).

[Synthesis Example 2] Synthesis of tris(2-acetoxyethyl)amine:

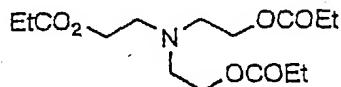


20 In 150 g of THF was dissolved 14.9 g of triethanolamine. To the resulting solution was added 40.4 g of triethylamine. After cooling over an ice bath, 27.5 g of acetyl chloride was 25 added dropwise at a temperature not exceeding 30°C. The

reaction mixture was heated to 75°C over an oil bath and heated for 2 hours to be refluxed. After completion of the maturation, the reaction mixture was allowed to cool down to room temperature. The reaction was terminated by the 5 addition of 100 g of water. To the organic layer thus separated was added 200 g of dichloromethane. The organic layer was washed with 100 g of water. The organic layer was concentrated under reduced pressure, followed by distillation under reduced pressure, whereby 20.6 g of the title tris(2-acetoxyethyl)amine was obtained (boiling point: 145 to 10 148°C/70 Pa, yield: 75%).

$^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta = 2.02(9\text{H},\text{s}), 2.80(6\text{H},\text{t}), 4.08(6\text{H},\text{t})$ .

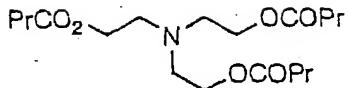
15 [Synthesis Example 3] Synthesis of tris(2-propionyloxyethyl)amine:



In a similar manner to Synthesis Example 2 except for the use of propionyl chloride instead of acetyl chloride, 20 tris(2-propionyloxyethyl)amine was synthesized (boiling point: 154 to 155°C/30 Pa, yield: 82%).

$^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta = 1.09(9\text{H},\text{t}), 2.29(6\text{H},\text{q}), 2.80(6\text{H},\text{t}), 4.08(6\text{H},\text{t})$ .

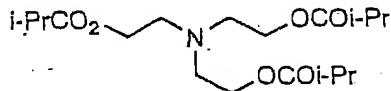
25 [Synthesis Example 4] Synthesis of tris(2-butyryloxyethyl)amine:



In a similar manner to Synthesis Example 2 except for the use of butyryl chloride instead of acetyl chloride, 5 tris(2-butyryloxyethyl)amine was synthesized (boiling point: 183 to 185°C/30 Pa, yield: 68%).

<sup>1</sup>H-NMR (300 MHz in CDCl<sub>3</sub>):  $\delta$  = 0.92(9H,t), 1.62(6H,m), 2.25(6H,t), 4.09(6H,t), 2.81(6H,t).

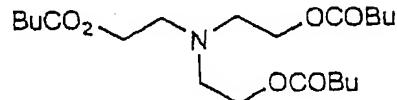
[Synthesis Example 5] Synthesis of tris(2-isobutyryloxyethyl)amine:



In a similar manner to Synthesis Example 2 except for the use of isobutyryl chloride instead of acetyl chloride, 15 tris(2-isobutyryloxyethylamine was synthesized (boiling point: 179 to 180°C/30Pa, yield: 87%).

<sup>1</sup>H-NMR (300 MHz in CDCl<sub>3</sub>):  $\delta$  = 1.13(18H,d), 2.51(3H,m), 2.82(6H,t), 4.09(6H,t).

[Synthesis Example 6] Synthesis of tris(2-valeryloxyethyl)amine:

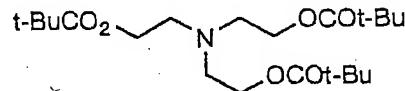


In a similar manner to Synthesis Example 2 except for 25 the use of valeryl chloride instead of acetyl chloride,

tris(2-valeryloxyethyl)amine was synthesized (boiling point: 190 to 193°C/30Pa, yield: 75%).

<sup>1</sup>H-NMR (300 MHz in CDCl<sub>3</sub>):  $\delta$  = 0.89(9H,t), 1.32(6H,m), 1.58(6H,m), 2.28(6H,t), 4.10(6H,t).

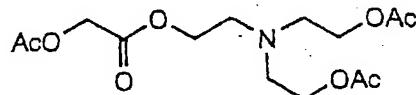
5 [Synthesis Example 7] Synthesis of tris(2-pivaloyloxyethyl)amine:



In a similar manner to Synthesis Example 2 except for the use of pivaloyl chloride instead of acetyl chloride, tris(2-pivaloyloxyethyl)amine was synthesized (boiling point: 183 to 185°C/30Pa, yield: 60%).

<sup>1</sup>H-NMR (300 MHz in CDCl<sub>3</sub>):  $\delta$  = 1.17(27H,s), 2.83(6H,t), 4.09(6H,t).

15 [Synthesis Example 8] Synthesis of N,N-bis(2-acetoxyethyl)-2-(acetoxyacetoxy)ethylamine:

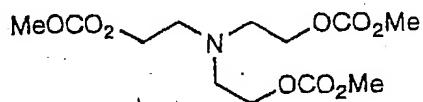


In 150 g of THF was dissolved 14.9 g of triethanolamine. To the resulting solution was added 40.4 g of triethylamine. The resulting mixture was cooled over ice bath and 15.7 g of acetyl chloride was added dropwise at a temperature not exceeding 30°C. The reaction mixture was heated over an oil bath to 75°C and heating was conducted for 2 hours under reflux. Then, 20.5 g of acetoxyacetyl chloride was added and

the resulting mixture was heated under reflux for further 2 hours. After the reaction mixture was allowed to be cooled down, 100 g of water was added to terminate the reaction. The organic layer thus separated was added with 200 g of dichloromethane, and then washed with 100 g of water. The 5 organic layer was concentrated under reduced pressure, followed by distillation under reduced pressure, whereby 18.3 g of N,N-bis(2-acetoxyethyl)-2-(acetoxycetoxo)ethylamine was obtained as the title compound (boiling point: 153 to 10 155°C/24 Pa, yield: 55%).

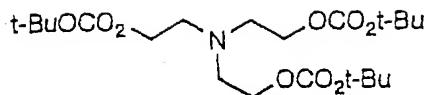
IR (thin film):  $\nu = 2958, 2837, 1740, 1425, 1373, 1238, 1198, 1082, 1043 \text{ cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta = 2.03$  (6H, s), 2.13 (3H, s), 2.75-2.85 (6H, m), 4.08 (4H, t,  $J=6.2\text{Hz}$ ), 4.19 (2H, t,  $J=6.0\text{Hz}$ ), 4.58 (2H, s).

15 [Synthesis Example 9] Synthesis of tris(2-methoxycarbonyloxyethyl)amine:



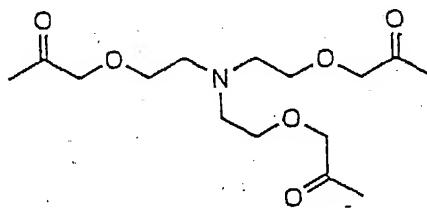
In a similar manner to Synthesis Example 2 except for 20 the use of methyl chloroformate instead of acetyl chloride, tris(2-methoxycarbonyloxyethyl)amine was synthesized.

[Synthesis Example 10] Synthesis of tris(2-tert-butoxycarbonyloxyethyl)amine:



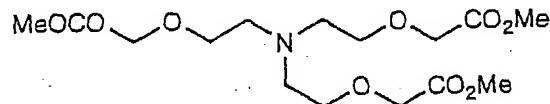
In a similar manner to Synthesis Example 2 except for the use of di-tert-butyl pyrocarboxylate instead of acetyl chloride, tris(2-tert-butoxycarbonyloxyethyl)amine was synthesized.

[Synthesis Example 11] Synthesis of tris[2(2-oxopropoxy)ethyl]amine:



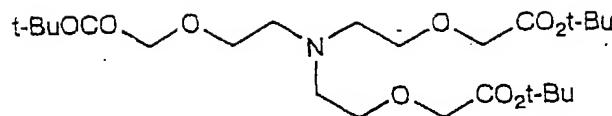
In a similar manner to Synthesis Example 2 except for the use of chloroacetone instead of acetyl chloride, tris[2-(2-oxopropoxy)ethyl]amine was synthesized.

[Synthesis Example 12] Synthesis of tris[2-(methoxycarbonylmethyl)oxyethyl]amine:



20 In a similar manner to Synthesis Example 2 except that methyl chloroacetate was used instead of acetyl chloride and 1.0 g of sodium iodide was added further, tris[2-(methoxycarbonylmethyl)oxyethyl]amine was synthesized.

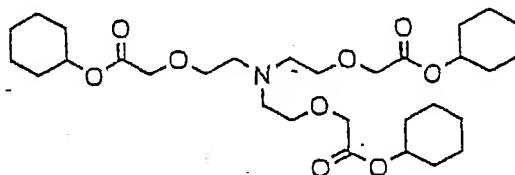
25 [Synthesis Example 13] Synthesis of tris[2-(tert-butoxycarbonylmethoxy)ethyl]amine:



In a similar manner to Synthesis Example 12 except for  
5 the use of tert-butyl chloroacetate instead of methyl  
chloroacetate, tris[2-(tert-  
butoxycarbonylmethyl)oxyethyl]amine was synthesized.

[Synthesis Example 14] Synthesis of tris[2-(cyclohexyloxycarbonylmethyloxy)ethyl]amine:

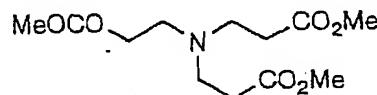
10



In a similar manner to Synthesis Example 12 except for  
15 the use of cyclohexyl chloroacetate instead of methyl  
chloroacetate, tris[2-  
(cyclohexyloxycarbonylmethyloxy)ethyl]amine was synthesized.

[Synthesis Example 15] Synthesis of tris(2-methoxycarbonylethyl)amine:

20



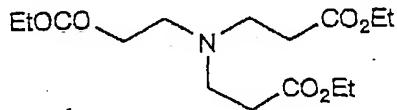
While vigorously stirring at 0°C, 5.00 g of 28% aqueous ammonia was added to 21.2 g of methyl acrylate. The resulting mixture was heated gradually to 25 °C over 20

hours. After concentration under reduced pressure, the residue was purified by distillation under reduced pressure, whereby tris(2-methoxycarbonylethyl)amine was obtained (boiling point: 143°C/70 Pa, yield: 76%),

5 IR (thin film):  $\nu$  = 2954, 2839, 1738, 1437, 1255, 1200, 1172 cm<sup>-1</sup>. <sup>1</sup>H-NMR (300 MHz in CDCl<sub>3</sub>):  $\delta$  = 2.41 (6H, t, J=7.7Hz), 2.73 (6H, t, J=7.7Hz), 3.63 (9H, s).

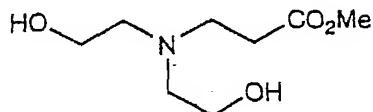
[Synthesis Example 16] Synthesis of tris(2-ethoxycarbonylethyl)amine:

10



In a similar manner to Synthesis Example 15 except for the use of ethyl acrylate instead of methyl acrylate, tris(2-ethoxycarbonylethyl)amine was synthesized.

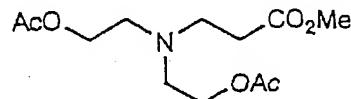
15 [Synthesis Example 17] Synthesis of N,N-bis(2-hydroxyethyl)-2-(methoxycarbonyl)ethylamine:



20 At 20 to 30°C, 9.01 g of methyl acrylate was added to 10.0 g of diethanolamine. The resulting mixture was allowed to stand for 20 hours. The reaction mixture was concentrated under reduced pressure, whereby N,N-bis(2-hydroxyethyl)-2-(methoxycarbonyl)ethylamine was obtained (yield: 98%).

IR (thin film):  $\nu = 3385$  (br.), 2953, 2881, 2833, 1734, 1439, 1404, 1363, 1259, 1198, 1035  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (270 MHz in  $\text{CDCl}_3$ ):  $\delta = 2.48$  (2H, t,  $J=6.3\text{Hz}$ ), 2.62 (4H, t,  $J=5.1\text{Hz}$ ), 2.83 (2H, t,  $J=6.3\text{Hz}$ ), 3.01 (1H, br.s), 3.58 (4H, t,  $J=5.1\text{Hz}$ , 5 3.67 (3H, s).

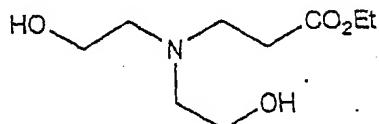
[Synthesis Example 18] Synthesis of N,N-bis(2-acetoxyethyl)-2-(methoxycarbonyl)ethylamine:



10 At 20 to 30°C, acetic anhydride was added to a mixture of 9.29 g of N,N-bis(2-hydroxyethyl)-2-(methoxycarbonyl)ethylamine, 12.8 g of triethylamine, 50 mg of 4-dimethylaminopyridine and 50 ml of THF and the resulting mixture was stirred for 20 hours. Water was added to 15 terminate the reaction. The reaction mixture was extracted with ethyl acetate. The organic layer was then washed with water and dried over anhydrous sodium sulfate. After concentration under reduced pressure, the residue was purified by distillation under reduced pressure, whereby 13.2 20 g of N,N-bis(2-acetoxyethyl)-2-(methoxycarbonyl)ethylamine was obtained (boiling point: 118 to 123°C/21Pa, yield: 99%).

IR (thin film):  $\nu = 2954, 2837, 1740, 1437, 1373, 1238, 1203, 1039$   $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (270 MHz in  $\text{CDCl}_3$ ):  $\delta = 2.02$  (6H, s), 2.42 (2H, t,  $J=7.1\text{Hz}$ ), 2.74 (4H, t,  $J=5.9\text{Hz}$ ), 2.87 (2H, t,  $J=7.1\text{Hz}$ , 25 3.64 (3H, s), 4.07 (4H, t,  $J=5.9\text{Hz}$ ).

[Synthesis Example 19] Synthesis of N,N-bis(2-hydroxyethyl)-2-(ethoxycarbonyl)ethylamine:

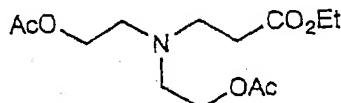


5

In a similar manner to Synthesis Example 17 except for the use of ethyl acrylate instead of methyl acrylate, N,N-bis(2-hydroxyethyl)-2-(ethoxycarbonyl)ethylamine was synthesized (yield: 94%).

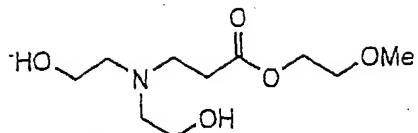
10 IR (thin film):  $\nu$  = 3400 (br.), 2951, 2875, 2829, 1732, 1462, 1400, 1371, 1255, 1190, 1039  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta$  = 1.24 (3H, t,  $J=7.2\text{Hz}$ ), 2.46 (2H, t,  $J=7.1\text{Hz}$ ), 2.60 (4H, m), 2.82 (2H, t,  $J=7.1\text{Hz}$ ), 3.17 (2H, br.s), 3.57 (4H, m), 4.12 (2H, q,  $J=7.2\text{Hz}$ ).

15 [Synthesis Example 20] Synthesis of N,N-bis(2-acetoxyethyl)-2-(ethoxycarbonyl)ethylamine:



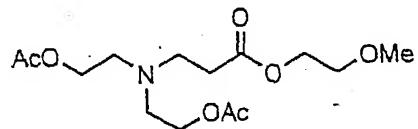
20 In a similar manner to Synthesis Example 18 except for the use of N,N-bis(2-hydroxyethyl)-2-(ethoxycarbonyl)ethylamine instead of N,N-bis(2-hydroxyethyl)-2-(methoxycarbonyl)ethylamine, N,N-bis(2-acetoxyethyl)-2-(ethoxycarbonyl)ethylamine was synthesized.

25 [Synthesis Example 21] Synthesis of N,N-bis(2-hydroxyethyl)-2-(2-methoxyethoxycarbonyl)ethylamine:



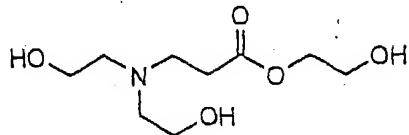
In a similar manner to Synthesis Example 17 except for  
 5 the use of 2-methoxyethyl acrylate instead of methyl  
 acrylate, N,N-bis(2-hydroxyethyl)-2-(2-  
 methoxyethoxycarbonyl)ethylamine was synthesized (yield:  
 97%).

[Synthesis Example 22] Synthesis of N,N-bis(2-acetoxyethyl)-  
 10 2-(2-methoxyethoxycarbonyl)ethylamine



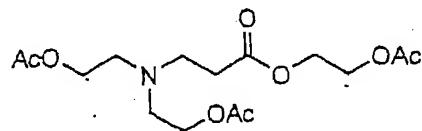
In a similar manner to Synthesis Example 18 except for  
 15 the use of N,N-bis(2-hydroxyethyl)-2-(2-  
 methoxyethoxycarbonyl)ethylamine instead of N,N-bis(2-  
 hydroxyethyl)-2-(methoxycarbonyl)ethylamine, N,N-bis(2-  
 acetoxyethyl)-2-(2-methoxyethoxycarbonyl)ethylamine was  
 synthesized (boiling point: 146 to 148°C/9.3Pa, yield: 86%).  
 20 IR (thin film):  $\nu$  = 2954, 2893, 2825, 1738, 1456, 1371,  
 1238, 1198, 1130, 1039  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta$  =  
 2.02(6H,s), 2.46(2H,t,J=7.1Hz), 2.74(4H,t,J=6.0Hz),  
 2.88(2H,t,J=7.1Hz), 3.36(3H,s), 3.56(2H,m),  
 4.08(4H,t,J=6.0Hz), 4.20(2H,m).  
 25 [Synthesis Example 23] Synthesis of N,N-bis(2-hydroxyethyl)-

2-(2-hydroxyethoxycarbonyl)ethylamine:



In a similar manner to Synthesis Example 17 except for  
5 the use of 2-hydroxyethyl acrylate instead of methyl  
acrylate, N,N-bis(2-hydroxyethyl)-2-(2-  
hydroxyethoxycarbonyl)ethylamine was synthesized (yield:  
99%).

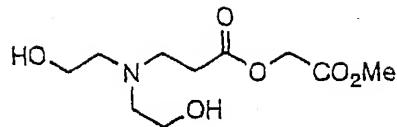
[Synthesis Example 24] Synthesis of N,N-bis(2-acetoxyethyl)-  
10 2-(2-acetoxyethoxycarbonyl)ethylamine:



In a similar manner to Synthesis Example 18 except for  
15 the use of N,N-bis(2-hydroxyethyl)-2-(2-  
hydroxyethoxycarbonyl)ethylamine instead of N,N-bis(2-  
hydroxyethyl)-2-(methoxycarbonyl)ethylamine, N,N-bis(2-  
acetoxyethyl)-2-(2-acetoxyethoxycarbonyl)ethylamine was  
synthesized (boiling point: 164 to 166°C/27Pa, yield: 96%).

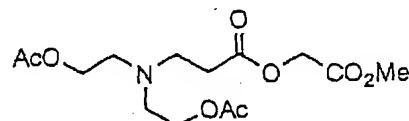
20 IR (thin film):  $\nu$  = 2960, 2837, 1740, 1443, 1375, 1236,  
1190, 1043  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta$  = 2.02 (6H, s),  
2.05 (3H, s), 2.45 (2H, t,  $J=6.9\text{Hz}$ ), 2.75 (4H, t,  $J=6.1\text{Hz}$ ),  
2.88 (2H, t,  $J=6.9\text{Hz}$ ), 4.08 (4H, t,  $J=6.1\text{Hz}$ ), 4.25 (4H, s).

[Synthesis Example 25] Synthesis of N,N-bis(2-hydroxyethyl)-  
25 2-[(methoxycarbonyl)methoxycarbonyl]ethylamine:



In a similar manner to Synthesis Example 17 except for  
 5 the use of methoxycarbonylmethyl acrylate instead of methyl acrylate, N,N-bis(2-hydroxyethyl)-2-[(methoxycarbonyl)-methoxycarbonyl]ethylamine was synthesized.

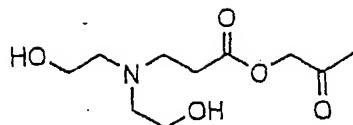
[Synthesis Example 26] Synthesis of N,N-bis(2-acetoxyethyl)-2-[(methoxycarbonyl)methoxycarbonyl]ethylamine:



10 In a similar manner to Synthesis Example 18 except for the use of N,N-bis(2-hydroxyethyl)-2-[(methoxycarbonyl)methoxycarbonyl]ethylamine instead of N,N-bis(2-hydroxyethyl)-2-(methoxycarbonyl)ethylamine, N,N-bis(2-acetoxyethyl)-2-[(methoxycarbonyl)methoxycarbonyl]ethylamine was synthesized (boiling point: 154 to 157°C/17Pa, yield: 15 60%).

IR (thin film):  $\nu = 2956, 2837, 1740, 1439, 1377, 1236, 20 1180, 1041 \text{ cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta = 2.02(6\text{H},s), 2.54(2\text{H},t,J=7.1\text{Hz}), 2.76(4\text{H},t,J=5.9\text{Hz}), 2.92(2\text{H},t,J=7.1\text{Hz}), 3.74(3\text{H},s), 4.09(4\text{H},t,J=5.9\text{Hz}), 4.59(2\text{H},s).$

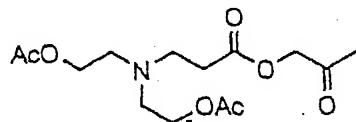
[Synthesis Example 27] Synthesis of N,N-bis(2-hydroxyethyl)-2-(2-oxopropoxycarbonyl)ethylamine:



In a similar manner to Synthesis Example 17 except for  
 5 the use of 2-oxopropyl acrylate instead of methyl acrylate,  
 N,N-bis(2-hydroxyethyl)-2-(2-oxopropoxycarbonyl)ethylamine  
 was synthesized (yield: 98%).

[Synthesis Example 28] Synthesis of N,N-bis(2-acetoxyethyl)-  
 2-(2-oxopropoxycarbonyl)ethylamine:

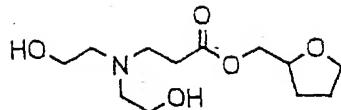
10



In a similar manner to Synthesis Example 18 except for  
 the use of N,N-bis(2-hydroxyethyl)-2-(2-  
 15 oxopropoxycarbonyl)ethylamine instead of N,N-bis(2-  
 hydroxyethyl)-2-(methoxycarbonyl)ethylamine, N,N-bis(2-  
 acetoxyethyl)-2-(2-oxopropoxycarbonyl)ethylamine was  
 synthesized (boiling point: 165°C/27Pa, yield: 85%).

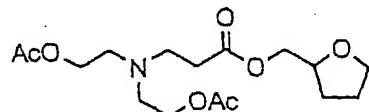
IR (thin film):  $\nu = 2960, 2837, 1736, 1421, 1373, 1238,$   
 20  $1174, 1041 \text{ cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta = 2.02(6\text{H},\text{s}),$   
 $2.13(3\text{H},\text{s}), 2.55(2\text{H},\text{t},J=7.1\text{Hz}), 2.76(4\text{H},\text{t},J=5.9\text{Hz}),$   
 $2.92(2\text{H},\text{t},J=7.1\text{Hz}), 4.08(4\text{H},\text{t},J=5.9\text{Hz}), 4.63(2\text{H},\text{s})$ .

[Synthesis Example 29] Synthesis of N,N-bis(2-hydroxyethyl)-  
 2-(tetrahydrofurfuryloxycarbonyl)ethylamine:



In a similar manner to Synthesis Example 17 except for the use of tetrahydrofurfuryl acrylate instead of methyl acrylate, N,N-bis(2-hydroxyethyl)-2-(tetrahydrofuryloxycarbonyl)ethylamine was synthesized (yield: 97%).

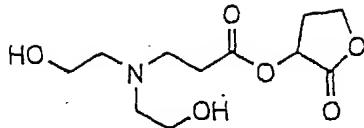
[Synthesis Example 30] Synthesis of N,N-bis(2-acetoxyethyl)-2-(tetrahydrofuryloxycarbonyl)ethylamine:



In a similar manner to Synthesis Example 18 except for the use of N,N-bis(2-hydroxyethyl)-2-(tetrahydrofuryloxycarbonyl)ethylamine instead of N,N-bis(2-hydroxyethyl)-2-(methoxycarbonyl)ethylamine, N,N-bis(2-acetoxyethyl)-2-(tetrahydrofuryloxycarbonyl)ethylamine was synthesized (boiling point: 165°C/20Pa, yield: 76%).

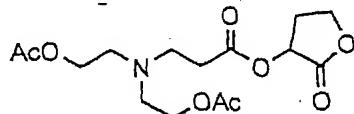
IR (thin film):  $\nu$  = 2958, 2873, 1740, 1450, 1371, 1238, 1193, 1039  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta$  = 0.56 (1H, m), 1.80-2.10 {10H, m including 2.02(6H, s)}, 2.47(2H, t,  $J$ =7.1Hz), 2.74(4H, t,  $J$ =6.0Hz), 2.88(2H, t,  $J$ =7.1Hz), 3.70-4.20 {9H, m including 4.06(4H, t,  $J$ =6.0H)}.

[Synthesis Example 31] Synthesis of N,N-bis(2-hydroxyethyl)-2-[(2-oxotetrahydrofuran-3-yl)oxycarbonyl]ethylamine:



In a similar manner to Synthesis Example 17 except for  
 5 the use of (2-oxotetrahydrofuran-3-yl) instead of methyl  
 acrylate, N,N-bis(2-hydroxyethyl)-2-[(2-oxotetrahydrofuran-3-  
 yl)oxycarbonyl]ethylamine was synthesized.

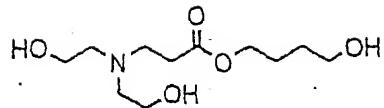
[Synthesis Example 32] Synthesis of N,N-bis(2-acetoxyethyl)-  
 2-[(2-oxotetrahydrofuran-3-yl)oxycarbonyl]ethylamine:



In a similar manner to Synthesis Example 18 except that  
 10 N,N-bis(2-hydroxyethyl)-2-[(2-oxotetrahydrofuran-3-  
 yl)oxycarbonyl]ethylamine was used instead of N,N-bis(2-  
 hydroxyethyl)-2-(methoxycarbonyl)ethylamine, and purification  
 15 was conducted by silica gel chromatography, N,N-bis(2-  
 acetoxyethyl)-2-[(2-oxotetrahydrofuran-3-  
 yl)oxycarbonyl]ethylamine was synthesized (yield: 70%).

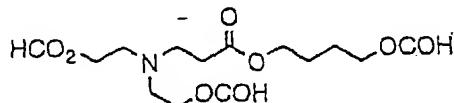
IR (thin film):  $\nu$  = 2962, 2837, 1792, 1743, 1668, 1456, 1437,  
 20 1373, 1240, 1196, 1095, 1043  $\text{cm}^{-1}$ .

[Synthesis Example 33] Synthesis of N,N-bis(2-hydroxyethyl)-  
 2-(4-hydroxybutoxycarbonyl)ethylamine



In a similar manner to Synthesis Example 17 except for the use of 4-hydroxybutyl acrylate instead of methyl acrylate, N,N-bis(2-hydroxyethyl)-2-(4-hydroxybutoxycarbonyl)ethylamine was synthesized.

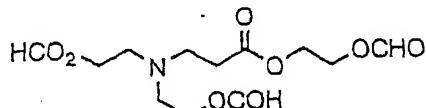
5 [Synthesis Example 34] Synthesis of N,N-bis(2-formyloxyethyl)-2-(4-formyloxybutoxycarbonyl)ethylamine:



In a similar manner to Synthesis Example 1 except for the use of N,N-bis(2-hydroxyethyl)-2-(4-hydroxybutoxycarbonyl)ethylamine instead of triethanolamine, N,N-bis(2-formyloxyethyl)-2-(4-formyloxybutoxycarbonyl)ethylamine was synthesized (yield: 15 70%).

20 IR (thin film):  $\nu = 2960, 2839, 1722, 1466, 1363, 1254, 1176, 1065 \text{ cm}^{-1}$ .  $^1\text{H-NMR}$  (270 MHz in  $\text{CDCl}_3$ ):  $\delta = 1.65-1.80(4\text{H},\text{m}), 2.44(2\text{H},\text{t},J=7.2\text{Hz}), 2.80(4\text{H},\text{t},J=5.8\text{Hz}), 2.89(2\text{H},\text{t},J=7.2\text{Hz}), 4.05-4.25(8\text{H},\text{m}), 8.03(2\text{H},\text{s}), 8.04(1\text{H},\text{s})$ .

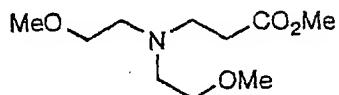
[Synthesis Example 35] Synthesis of N,N-bis(2-formyloxyethyl)-2-(2-formyloxyethoxycarbonyl)ethylamine:



In a similar manner to Synthesis Example 1 except for the use of N,N-bis(2-hydroxyethyl)-2-(2-hydroxyethoxycarbonyl)ethylamine instead of triethanolamine, N,N-bis(2-formyloxyethyl)-2-(2-formyloxyethoxycarbonyl)ethylamine was synthesized (yield: 5 40%).

IR (thin film):  $\nu$  = 2956, 2839, 1722, 1456, 1275, 1254, 1173, 1061  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (270 MHz in  $\text{CDCl}_3$ ):  $\delta$  = 2.47 (2H, t,  $J=7.0\text{Hz}$ ), 2.80 (4H, t,  $J=5.9\text{Hz}$ ), 2.90 (2H, t,  $J=7.0\text{Hz}$ ), 4.19 (4H, t,  $J=5.9\text{Hz}$ ), 10 4.25-4.40 (4H, m), 8.03 (2H, s), 8.06 (1H, s).

[Synthesis Example 36] Synthesis of N,N-bis(2-methoxyethyl)-2-(methoxycarbonyl)ethylamine:

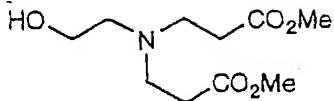


15 In a similar manner to Synthesis Example 17 except that bis(2-methoxyethyl)amine was used instead of diethanolamine, and purification was conducted by distillation under reduced pressure, N,N-bis(2-methoxyethyl)-2-(methoxycarbonyl)ethylamine was synthesized (boiling point: 20 71 to 75  $^{\circ}\text{C}$ /27Pa, yield: 93%).

IR (thin film):  $\nu$  = 2951, 2927, 2877, 2818, 1740, 1437, 1254, 1198, 1119  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (270 MHz in  $\text{CDCl}_3$ ):  $\delta$  = 2.46 (2H, t,  $J=7.3\text{Hz}$ ), 2.69 (4H, t,  $J=6.0\text{Hz}$ ), 2.89 (2H, t,  $J=7.3\text{Hz}$ ), 3.31 (6H, s), 3.43 (4H, t,  $J=6.0\text{Hz}$ ), 3.64 (3H, s).

25 [Synthesis Example 37] Synthesis of N-(2-hydroxyethyl)bis[2-

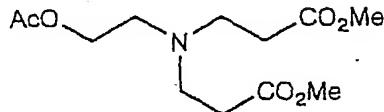
(methoxycarbonyl)ethyl]amine :



At 20 to 30°C, 15.5 g of methyl acrylate was added to  
5 5.00 g of 2-aminoethanol. The resulting mixture was allowed  
to stand for 20 hours, followed by concentration under  
reduced pressure, whereby 19.2 g of N-(2-hydroxyethyl)bis[2-  
(methoxycarbonyl)ethyl]amine was obtained (at a quantitative  
yield).

10 IR (thin film):  $\nu$  = 3467(br.), 2954, 2839, 1738, 1439, 1361,  
1257, 1201, 1176, 1038  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta$  =  
2.43(4H, t,  $J=6.8\text{Hz}$ ), 2.55(2H, m), 2.75(4H, t,  $J=6.8\text{Hz}$ ),  
2.93(1H, br. s), 3.54(2H, m), 3.64(6H, s).

[Synthesis Example 38] Synthesis of N-(2-acetoxyethyl)bis[2-  
15 (methoxycarbonyl)ethyl]amine :

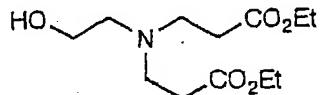


At 20 to 30°C, 2.56 g of acetic anhydride was added to a  
20 mixture of 5.00 g of N-(2-hydroxyethyl)bis[2-  
(methoxycarbonyl)ethyl]amine, 2.97 g of triethylamine, 50 mg  
of 4-dimethylaminopyridine and 25 g of THF. The resulting  
mixture was stirred for 5 hours. Water was added to  
terminate the reaction. The reaction mixture was then  
25 extracted with ethyl acetate. The organic layer was washed

with water and dried over anhydrous sodium sulfate. After concentration under reduced pressure, the residue was purified by distillation under reduced pressure, whereby 5.30 g of N-(2-acetoxyethyl)bis[2-(methoxycarbonyl)ethyl]amine was obtained (boiling point: 120°C/15Pa, yield: 92%).

5 IR (thin film):  $\nu$  = 2954, 2839, 1740, 1439, 1373, 1238, 1200, 1176, 1039  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta$  = 2.01 (3H, s), 2.41 (4H, t,  $J$ =6.9Hz), 2.67 (2H, t,  $J$ =6.0Hz), 2.79 (4H, t,  $J$ =6.9Hz), 3.63 (6H, s), 4.06 (2H, t,  $J$ =6.0Hz).

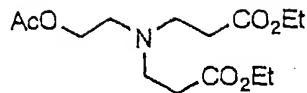
10 [Synthesis Example 39] Synthesis of N-(2-hydroxyethyl)bis[2-(ethoxycarbonyl)ethyl]amine:



15 In a similar manner to Synthesis Example 37 except for the use of ethyl acrylate instead of methyl acrylate, N-(2-hydroxyethyl)bis[2-(ethoxycarbonyl)ethyl]amine was synthesized (yield: 82%).

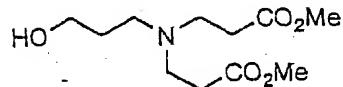
20 IR (thin film):  $\nu$  = 3502(br.), 2981, 2829, 1734, 1463, 1371, 1346, 1255, 1184, 1039  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta$  = 1.23 (6H, t,  $J$ =7.1Hz), 2.42 (4H, t,  $J$ =6.7Hz), 2.57 (2H, m), 2.77 (4H, t,  $J$ =6.7Hz), 2.96 (1H, m), 3.55 (2H, m), 4.10 (4H, q,  $J$ =7.1Hz).

[Synthesis Example 40] Synthesis of N-(2-acetoxyethyl)bis[2-(ethoxycarbonyl)ethyl]amine:



In a similar manner to Synthesis Example 38 except for the use of N-(2-hydroxyethyl)bis[2-(ethoxycarbonyl)ethyl]amine instead of N-(2-hydroxyethyl)bis[2-(methoxycarbonyl)ethyl]amine, N-(2-acetoxyethyl)bis[2-(ethoxycarbonyl)ethyl]amine was synthesized.

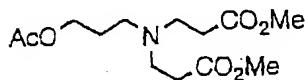
[Synthesis Example 41] Synthesis of N-(3-hydroxy-1-propyl)bis[2-(methoxycarbonyl)ethyl]amine:



In a similar manner to Synthesis Example 37 except for the use of 3-amino-1-propanol instead of 2-aminoethanol, N-(3-hydroxy-1-propyl)bis[2-(methoxycarbonyl)ethyl]amine was synthesized (at a quantitative yield).

IR (thin film):  $\nu = 3435(\text{br.}), 2953, 2841, 1738, 1439, 1358, 1331, 1255, 1200, 1174, 1134, 1051 \text{ cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta = 1.66(2\text{H},\text{m}), 2.47(4\text{H},\text{t},J=7.0\text{Hz}), 2.60(2\text{H},\text{m}), 2.76(4\text{H},\text{t},J=7.0\text{Hz}), 3.65(9\text{H},\text{s}), 3.68(2\text{H},\text{m}), 3.83(1\text{H},\text{br.s})$ .

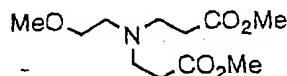
[Synthesis Example 42] Synthesis of N-(3-acetoxy-1-propyl)bis[2-(methoxycarbonyl)ethyl]amine:



In a similar manner to Synthesis Example 38 except for

the use of N-(3-hydroxy-1-propyl)bis[2-(methoxycarbonyl)ethyl]amine instead of N-(2-hydroxyethyl)bis[2-(methoxycarbonyl)ethyl]amine, N-(3-acetoxy-1-propyl)bis[2-(methoxycarbonyl)ethyl]amine was  
5 synthesized.

[Synthesis Example 43] Synthesis of N-(2-methoxyethyl)bis[2-(methoxycarbonyl)ethyl]amine:



10 In a similar manner to Synthesis Example 37 except for the use of 2-methoxyethylamine instead of 2-aminoethanol, N-(2-methoxyethyl)bis[2-(methoxycarbonyl)ethyl]amine was synthesized (boiling point: 81 to 85°C/27Pa, yield: 96%).

15 IR (thin film):  $\nu$  = 2953, 2839, 1740, 1437, 1255, 1200, 1176, 1119  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (270 MHz in  $\text{CDCl}_3$ ):  $\delta$  = 2.44 (4H, t,  $J=7.2\text{Hz}$ ), 2.63 (2H, t,  $J=6.1\text{Hz}$ ), 2.81 (4H, t,  $J=7.2\text{Hz}$ ), 3.31 (3H, s), 3.41 (2H, t,  $J=6.1\text{Hz}$ ), 3.64 (6H, s).

[Synthesis Example 44] Synthesis of N-butylbis[2-(methoxycarbonyl)ethyl]amine:



20 At 20 to 30°C, 12.9 g of methyl acrylate was added to 5.00 g of n-butylamine. The resulting mixture was allowed to stand for 20 hours, followed by concentration under reduced pressure, whereby 16.7 g of N-butylbis[2-

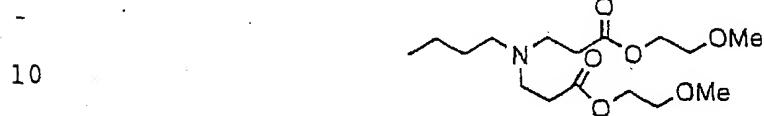
(methoxycarbonyl)ethyl]amine was obtained (at a quantitative yield).

IR (thin film):  $\nu = 2954, 1740, 1437, 1254, 1198, 1174 \text{ cm}^{-1}$ .

$^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta = 0.87(3\text{H},\text{t},J=7.2\text{Hz}), 1.15-$

5 1.45(4H,m), 2.37(2H,t,J=7.4Hz), 2.41(4H,t,J=7.1Hz),  
2.73(4H,t,J=7.1Hz), 3.64(6H,s).

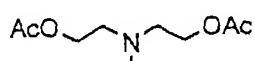
[Synthesis Example 45] Synthesis of N-butylbis[2-(2-methoxyethoxycarbonyl)ethyl]amine:



In a similar manner to Synthesis Example 44 except for the use of 2-methoxyethyl acrylate instead of methyl acrylate, N-butylbis[2-(2-methoxyethoxycarbonyl)ethyl]amine 15 was synthesized (boiling point: 156°C/28 Pa, yield: 90%).

IR (thin film):  $\nu = 2956, 2931, 2875, 2819, 1736, 1456, 1375, 1252, 1182, 1130, 1036 \text{ cm}^{-1}$ .  $^1\text{H-NMR}$  (300 MHz in  $\text{CDCl}_3$ ):  $\delta = 0.86(3\text{H},\text{t},J=7.2\text{Hz}), 1.15-1.45(4\text{H},\text{m}), 2.37(2\text{H},\text{t},J=7.1\text{Hz}), 2.46(4\text{H},\text{t},J=7.4\text{Hz}), 2.74(4\text{H},\text{t},J=7.4\text{Hz}), 3.35(6\text{H},\text{s}), 3.56(4\text{H},\text{m}), 4.19(4\text{H},\text{m})$ .

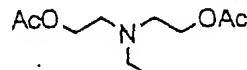
[Synthesis Example 46] Synthesis of N-methylbis(2-acetoxyethyl)amine:



In a similar manner to Synthesis Example 2 except for 25 the use of N-methyldiethanolamine instead of triethanolamine,

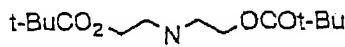
N-methylbis(2-acetoxyethyl)amine was synthesized.

[Synthesis Example 47] Synthesis of N-ethylbis(2-acetoxyethyl)amine :



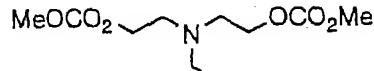
In a similar manner to Synthesis Example 2 except for the use of N-ethyldiethanolamine instead of triethanolamine, N-ethylbis(2-acetoxyethyl)amine was synthesized.

10 [Synthesis Example 48] Synthesis of N-methylbis(2-pivaloyloxyethyl)amine :



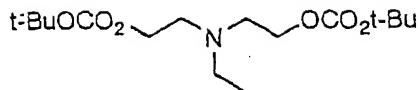
In a similar manner to Synthesis Example 7 except for the use of N-methyldiethanolamine instead of triethanolamine, N-methylbis(2-pivaloyloxyethyl)amine was synthesized.

15 [Synthesis Example 49] Synthesis of N-ethylbis[2-(methoxycarbonyloxy)ethyl]amine:



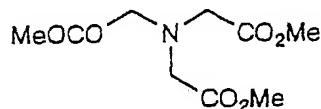
20 In a similar manner to Synthesis Example 9 except for the use of N-ethyldiethanolamine instead of triethanolamine, N-ethylbis[2-(methoxycarbonyloxy)ethyl]amine was synthesized.

[Synthesis Example 50] Synthesis of N-ethylbis[2-(tert-butoxycarbonyloxy)ethyl]amine :



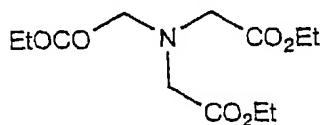
In a similar manner to Synthesis Example 10 except for the use of N-methyldiethanolamine instead of triethanolamine, 5 N-ethylbis[2-(tert-butoxycarbonyloxy)ethyl]amine was synthesized.

[Synthesis Example 51] Synthesis of tris(methoxycarbonylmethyl)amine:



At 0 to 10°C, 29.4 g of methyl chloroacetate, and then 70 g of a saturated aqueous solution of potassium carbonate were added to a mixture of 5.00 g of 28% aqueous ammonia and 50 g of THF. The resulting mixture was stirred at room 15 temperature for 20 hours. The reaction mixture was then extracted with diethyl ether, followed by washing with water, drying over anhydrous sodium sulfate and concentration under reduced pressure, whereby tris(methoxycarbonylmethyl)amine was obtained.

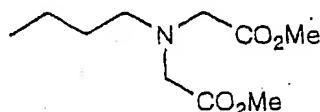
20 [Synthesis Example 52] Synthesis of tris(ethoxycarbonylmethyl)amine:



In a similar manner to Synthesis Example 51 except for 25

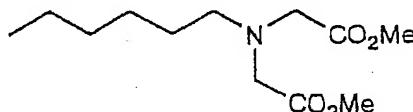
the use of ethyl chloroacetate instead of methyl chloroacetate, tris(ethoxycarbonylmethyl)amine was synthesized.

[Synthesis Example 53] Synthesis of N-  
5 butylbis(methoxycarbonylmethyl)amine:



In a similar manner to Synthesis Example 51 except for  
10 the use of n-butylamine instead of aqueous ammonia, N-  
butylbis(methoxycarbonylmethyl)amine was synthesized.

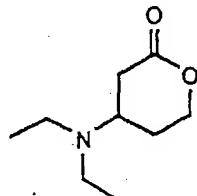
[Synthesis Example 54] Synthesis of N-  
hexylbis(methoxycarbonylmethyl)amine:



15

In a similar manner to Synthesis Example 51 except for  
the use of n-hexylamine instead of aqueous ammonia, N-  
hexylbis(methoxycarbonylmethyl)amine was synthesized.

[Synthesis Example 55] Synthesis of  $\beta$ -(diethylamino)- $\delta$ -  
20 valerolactone:



At 20 to 30°C, 4.00 g of diethylamine was added to 5.10  
25 g of 5,6-dihydro-2H-pyran-2-one. The resulting mixture was

allowed to stand for 20 hours, followed by concentration under reduced pressure, whereby  $\beta$ -(diethylamino)- $\delta$ -valerolactone was obtained.

[Example 1]

5 A resist solution was prepared by adding, at ratios as shown in the column of Example 1 of Table 1, the polymer, acid generator, base, dissolution inhibitor and crosslinker to 600 parts by weight of a 70:30 (weight ratio) mixed solvent of propylene glycol monomethyl ether acetate (PGMEA) 10 and ethyl lactate (EL) to dissolve the former in the latter, followed by filtration through a 0.1- $\mu\text{m}$  Teflon® filter.

[Examples 2 to 63, Comparative Examples 1 to 12]

In Examples 2 to 63, and Comparative Examples 1 to 12, resist solutions were prepared as in Example 1 under the 15 conditions as shown in Tables 1 to 5. The structure of each of Amines 1 to 55 and the structure, weight-average molecular weight (Mw) and its ratio to the number-average molecular weight (Mn) of each of Polymers 1 to 8 will be shown later.

Evaluation of a resist

20 In each of Examples 2 to 63 and Comparative Examples 1 to 12, a resist solution was prepared by adding the polymer, acid generator, base, dissolution inhibitor and crosslinker to a 70:30 (weight ratio) solvent mixture of propylene glycol monomethyl ether acetate (PGMEA) and ethyl lactate (EL) to 25 dissolve the former in the latter, followed by filtration

through a 0.1- $\mu\text{m}$  Teflon® filter.

The resist solution thus obtained was applied by spin coating onto a silicon wafer which has thereover a film of DUV-30 (trade name; product of Nissan Chemical) in the 5 thickness of 55 nm and has a reflectance suppressed to 1% or less by a KrF light (248 nm). The resulting silicon wafer was baked on a hot plate at 100°C for 90 seconds, whereby the thickness of the resist was adjusted to 550 nm.

The resulting resist was exposed to light of an excimer 10 laser stepper ("NSR-S202A", product of Nikon K.K., NA-0.5,  $\sigma$ : 0.75, 2/3 zonal light) at varied exposures and focuses. Immediately after exposure, the resist was baked at 110°C for 90 seconds and then developed for 60 seconds in a 2.38 wt% aqueous solution of tetramethylammonium hydroxide, whereby a 15 pattern was formed.

The resist pattern thus formed was evaluated in the following manner.

Provided that the exposure quantity with which a 0.16- $\mu\text{m}$  line-and-space pattern is resolved at 1:1 is the optimum 20 exposure (E<sub>op</sub>) which serves as an index of resist sensitivity, the "focus margin" at this time is determined. The definition for the focus margin is such that it is free from a decrease in the pattern film and its size falls within 0.16  $\mu\text{m} \pm 10\%$ .

25 Results are shown in Tables 1 to 5.

Table 1

Example	Polymer (part by weight)	Acid generator (part by weight)	Base (part by weight)	Dissolution inhibitor or crosslinker (part by weight)	Sensitivity (mJ/cm <sup>2</sup> )	Focus margin (μm)
1	1 (100)	PAG2 (2)	Amine 1 (0.1)	-	30	0.8
2	1 (100)	PAG2 (2)	Amine 2 (0.1)	-	35	0.9
3	1 (100)	PAG2 (2)	Amine 3 (0.1)	-	31	1.0
4	1 (100)	PAG2 (2)	Amine 4 (0.12)	-	30	1.0
5	1 (100)	PAG2 (2)	Amine 5 (0.12)	-	33	1.0
6	1 (100)	PAG2 (2)	Amine 6 (0.12)	-	28	1.0
7	1 (100)	PAG2 (2)	Amine 7 (0.1)	-	32	1.0
8	1 (100)	PAG2 (2)	Amine 8 (0.12)	-	39	1.0
9	1 (100)	PAG2 (2)	Amine 9 (0.16)	-	38	1.0
10	1 (100)	PAG2 (2)	Amine 10 (0.16)	-	36	1.0
11	2 (100)	PAG2 (2)	Amine 1 (0.1)	-	35	0.6
12	3 (100)	PAG2 (2)	Amine 1 (0.1)	-	31	1.1
13	4 (100)	PAG2 (2)	Amine 1 (0.1)	Crosslinker (15)	38	0.8
14	5 (100)	PAG1 (2)	Amine 1 (0.1)	-	33	0.8
15	6 (100)	PAG1 (2)	Amine 1 (0.1)	-	46	1.0
16	7 (100)	PAG1 (2)	Amine 1 (0.1)	-	48	1.0
17	8 (100)	PAG1 (2)	Amine 1 (0.1)	-	42	1.0
18	2 (100)	PAG2 (2)	Amine 1 (0.1)	DRI (20)	31	0.8

Table 2

Example	Polymer (part by weight)	Acid generator (part by weight)	Base (part by weight)	Dissolution inhibitor or crosslinker (part by weight)	Sensitivity (mJ/cm <sup>2</sup> )	Focus margin ( $\mu$ m)
19	1 (100)	PAG2 (2)	Amine 11 (0.1)	-	42	1.2
20	1 (100)	PAG2 (2)	Amine 12 (0.11)	-	40	1.0
21	1 (100)	PAG2 (2)	Amine 13 (0.12)	-	39	1.0
22	1 (100)	PAG2 (2)	Amine 14 (0.12)	-	38	1.0
23	1 (100)	PAG2 (2)	Amine 15 (0.13)	-	37	1.0
24	1 (100)	PAG2 (2)	Amine 16 (0.13)	-	36	0.9
25	1 (100)	PAG2 (2)	Amine 17 (0.12)	-	36	1.1
26	1 (100)	PAG2 (2)	Amine 18 (0.11)	-	38	1.0
27	1 (100)	PAG2 (2)	Amine 19 (0.09)	-	38	0.9
28	1 (100)	PAG2 (2)	Amine 20 (0.1)	-	38	1.0
29	1 (100)	PAG2 (2)	Amine 21 (0.09)	-	40	0.9
30	1 (100)	PAG2 (2)	Amine 22 (0.12)	-	38	1.0
31	1 (100)	PAG2 (2)	Amine 23 (0.1)	-	40	0.9
32	1 (100)	PAG2 (2)	Amine 24 (0.12)	-	38	1.0
33	1 (100)	PAG2 (2)	Amine 25 (0.1)	-	37	0.9
34	1 (100)	PAG2 (2)	Amine 26 (0.12)	-	36	1.2
35	1 (100)	PAG2 (2)	Amine 27 (0.13)	-	34	1.0
36	1 (100)	PAG2 (2)	Amine 28 (0.12)	-	36	1.1

Table 3

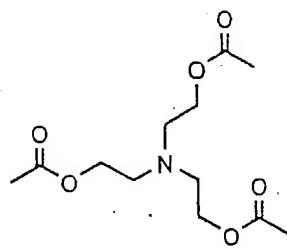
Example	Polymer (part by weight)	Acid generator (part by weight)	Base (part by weight)	Dissolution inhibitor or crosslinker (part by weight)	Sensitivity (mJ/cm <sup>2</sup> )	Focus margin ( $\mu$ m)
37	1 (100)	PAG2 (2)	Amine 29 (0.12)	-	37	1.0
38	1 (100)	PAG2 (2)	Amine 30 (0.16)	-	40	1.0
39	1 (100)	PAG2 (2)	Amine 31 (0.11)	-	36	0.9
40	1 (100)	PAG2 (2)	Amine 32 (0.13)	-	40	1.0
41	1 (100)	PAG2 (2)	Amine 33 (0.11)	-	37	0.9
42	1 (100)	PAG2 (2)	Amine 34 (0.14)	-	39	1.0
43	1 (100)	PAG2 (2)	Amine 35 (0.1)	-	34	0.9
44	1 (100)	PAG2 (2)	Amine 36 (0.12)	-	35	1.1
45	1 (100)	PAG2 (2)	Amine 37 (0.1)	-	40	1.2
46	1 (100)	PAG2 (2)	Amine 38 (0.1)	-	41	1.0
47	1 (100)	PAG2 (2)	Amine 39 (0.1)	-	38	1.0
48	1 (100)	PAG2 (2)	Amine 40 (0.1)	-	38	1.0
49	1 (100)	PAG2 (2)	Amine 41 (0.11)	-	41	1.0
50	1 (100)	PAG2 (2)	Amine 42 (0.1)	-	40	1.0
51	1 (100)	PAG2 (2)	Amine 43 (0.1)	-	38	1.0
52	1 (100)	PAG2 (2)	Amine 44 (0.1)	-	37	1.0
53	1 (100)	PAG2 (2)	Amine 45 (0.1)	-	37	0.9
54	1 (100)	PAG2 (2)	Amine 46 (0.12)	-	38	0.9

Table 4

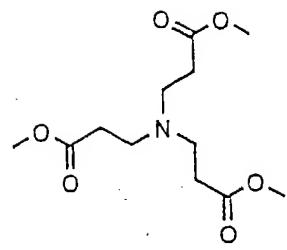
Example	Polymer (part by weight)	Acid generator (part by weight)	Base (part by weight)	Dissolution inhibitor or crosslinker (part by weight)	Sensitivity (mJ/cm <sup>2</sup> )	Focus margin (μm)
55	1 (100)	PAG2 (2)	Amine 47 (0.09)	-	34	0.9
56	1 (100)	PAG2 (2)	Amine 48 (0.11)	-	35	0.8
57	1 (100)	PAG2 (2)	Amine 49 (0.11)	-	34	0.9
58	1 (100)	PAG2 (2)	Amine 50 (0.12)	-	36	0.8
59	1 (100)	PAG2 (2)	Amine 51 (0.12)	-	34	0.8
60	1 (100)	PAG2 (2)	Amine 52 (0.1)	-	40	1.0
61	1 (100)	PAG2 (2)	Amine 53 (0.11)	-	39	1.0
62	1 (100)	PAG2 (2)	Amine 54 (0.1)	-	40	0.9
63	1 (100)	PAG2 (2)	Amine 55 (0.11)	-	40	0.9

Table 5

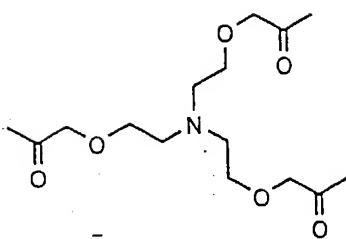
Comparative Example	Polymer (part by weight)	Acid generator (part by weight)	Base (part by weight)	Dissolution inhibitor or crosslinker (part by weight)	Sensitivity (mJ/cm <sup>2</sup> )	Focus margin (μm)
1	1 (100)	PAG2 (2)	-	-	5	0
2	1 (100)	PAG2 (2)	Proton sponge (0.2)	-	30	0.4
3	1 (100)	PAG2 (2)	DBN (0.1)	-	25	0.4
4	1 (100)	PAG2 (2)	DBU (0.1)	-	28	0.5
5	2 (100)	PAG2 (2)	DBN (0.1)	-	35	0.2
6	3 (100)	PAG2 (2)	DBN (0.1)	-	31	0.5
7	4 (100)	PAG2 (2)	DBN (0.1)	Crosslinker (15)	38	0.3
8	5 (100)	PAG1 (2)	DBN (0.1)	-	33	0.3
9	6 (100)	PAG1 (2)	DBN (0.1)	-	46	0.6
10	7 (100)	PAG1 (2)	DBN (0.1)	-	48	0.6
11	8 (100)	PAG1 (2)	DBN (0.1)	-	42	0.6
12	2 (100)	PAG2 (2)	DBN (0.1)	DRI (20)	31	0.4



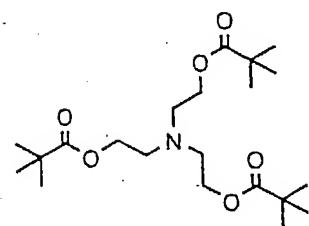
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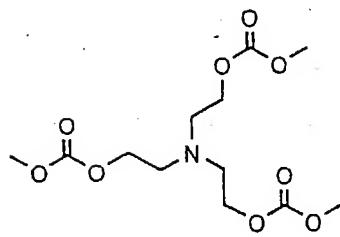
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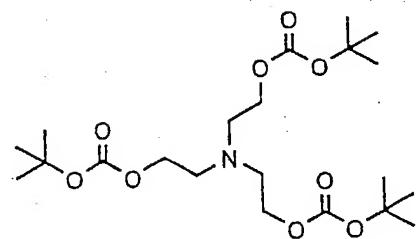
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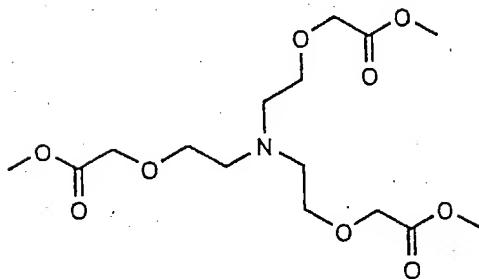
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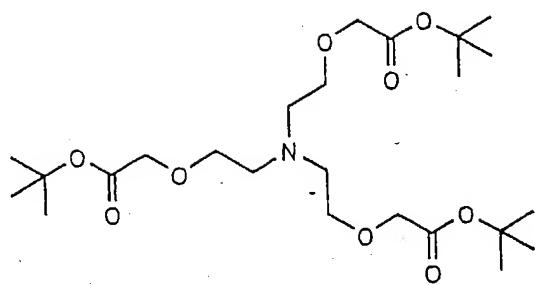
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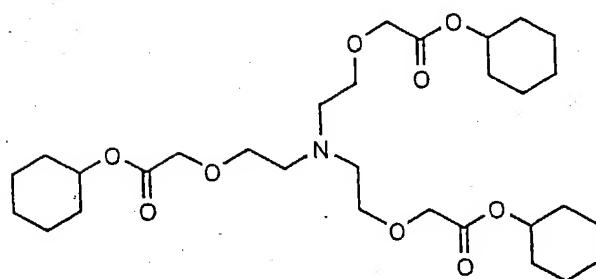
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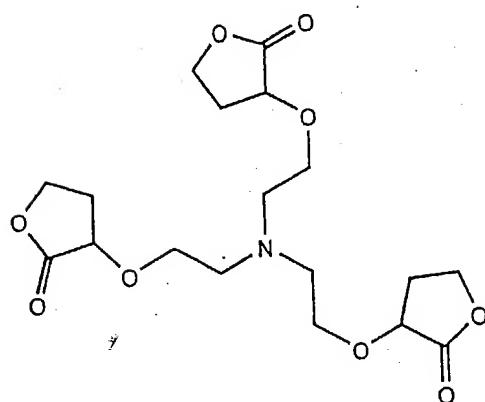
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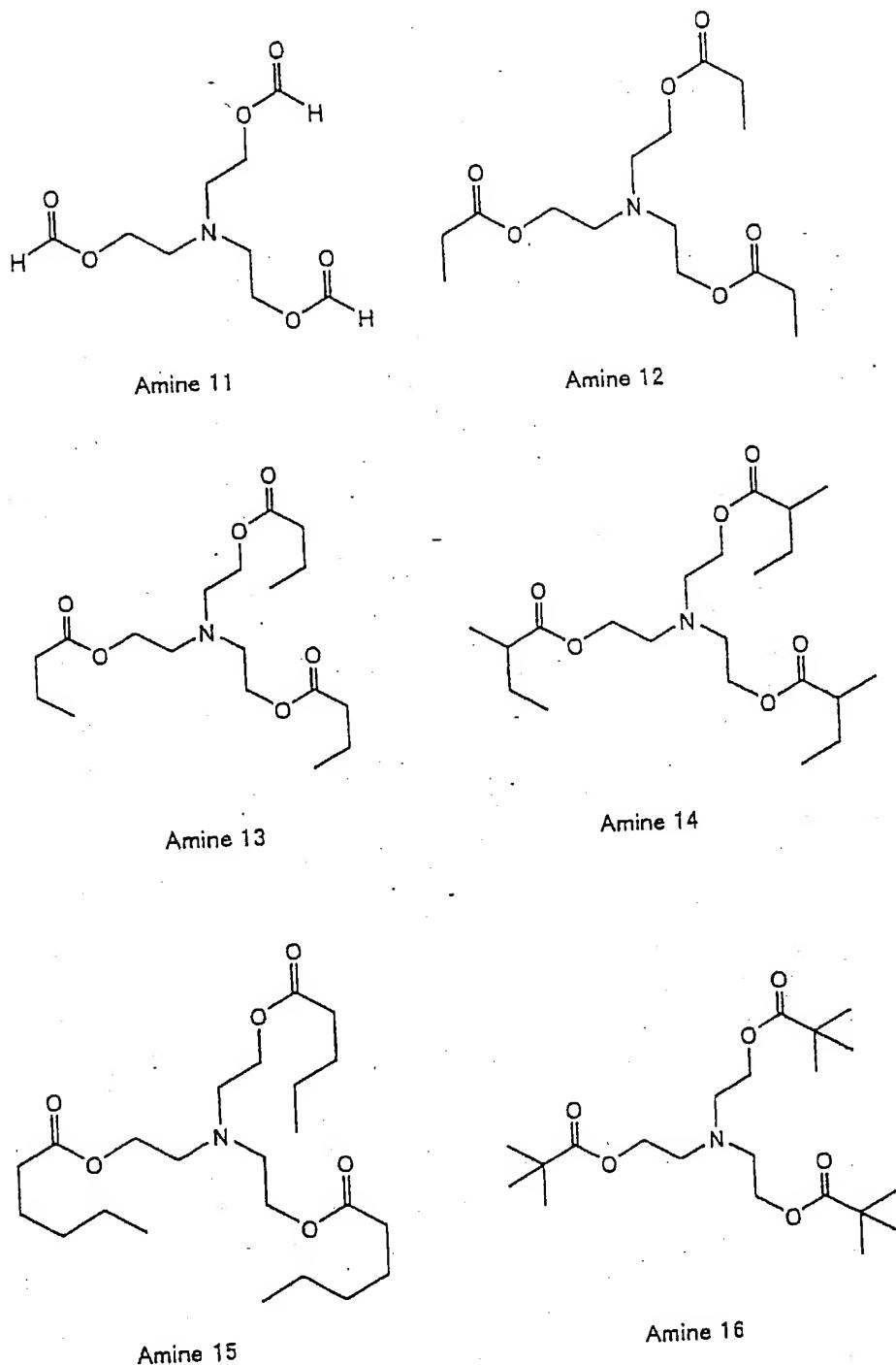
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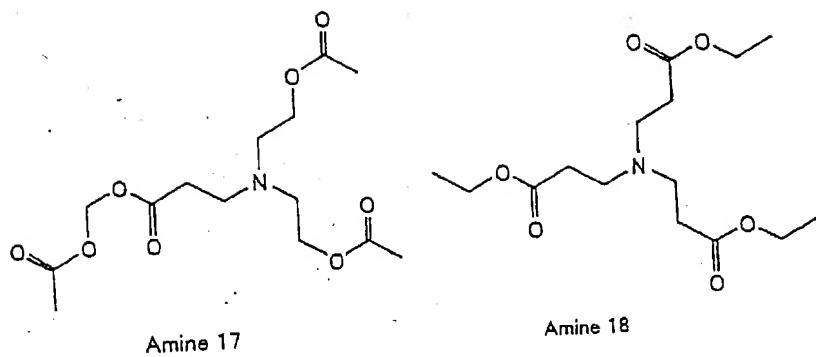


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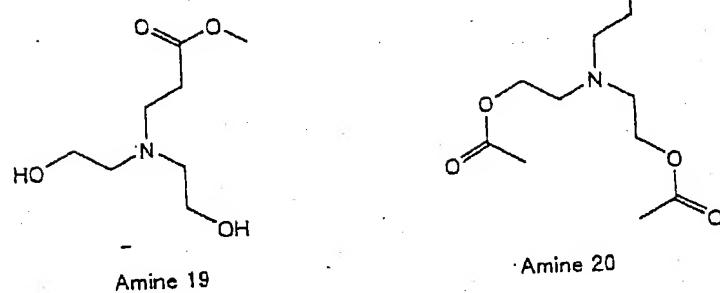
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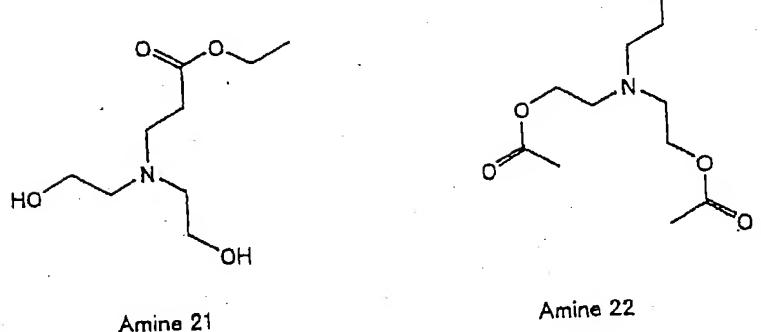
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Amine 18



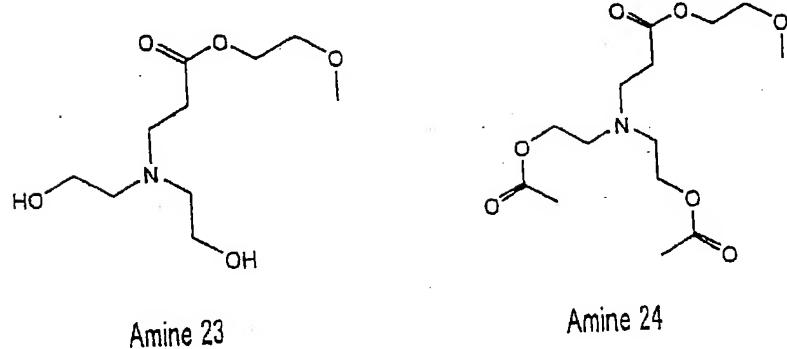
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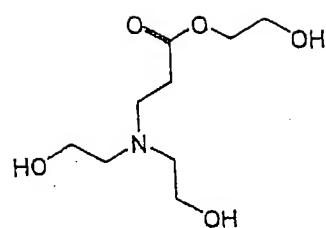
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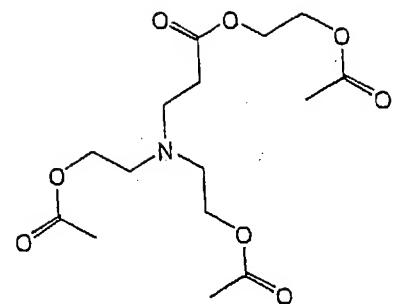


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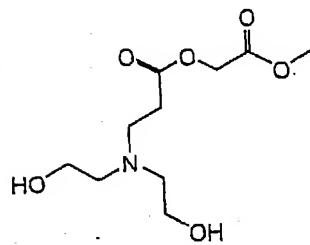
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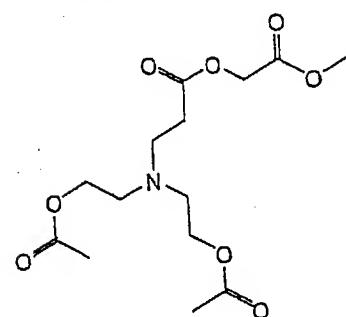
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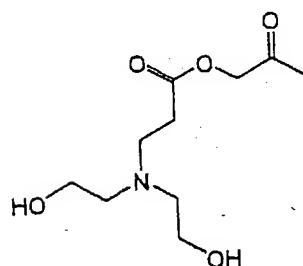
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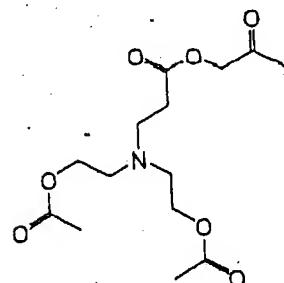
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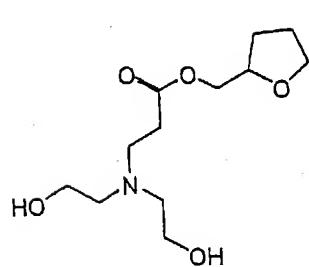
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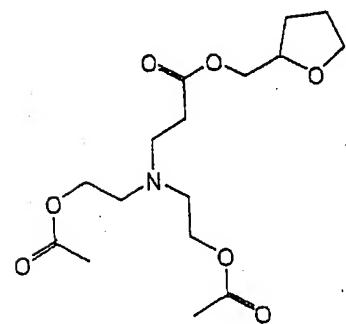
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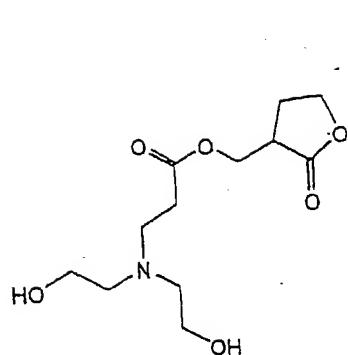
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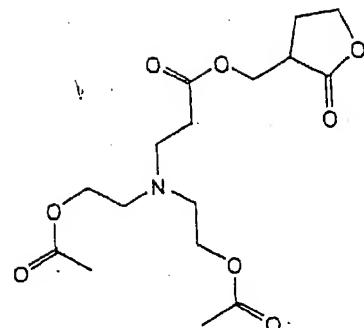
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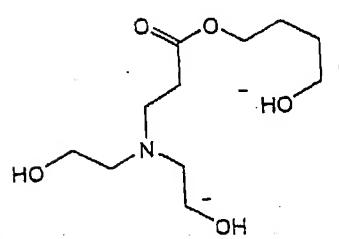
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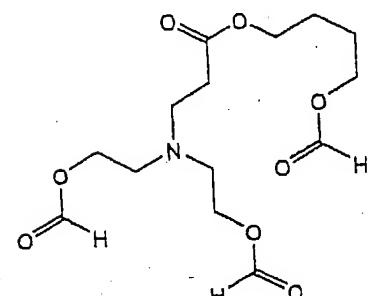
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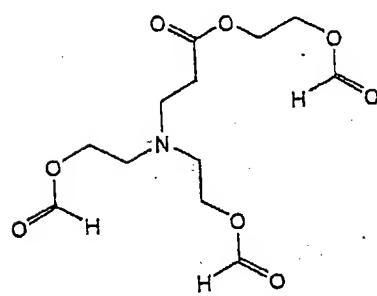
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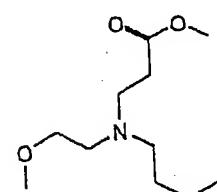
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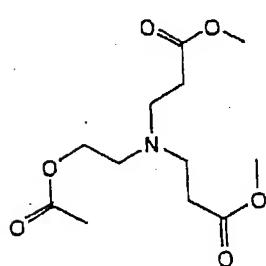
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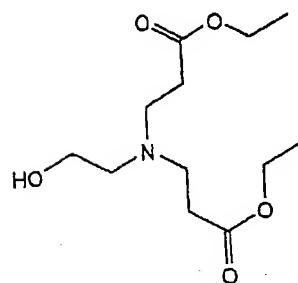
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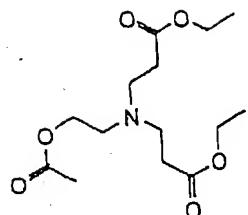
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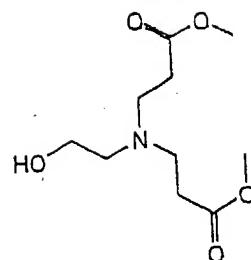
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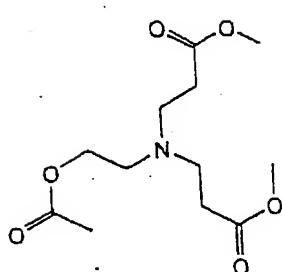
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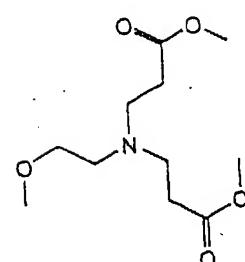
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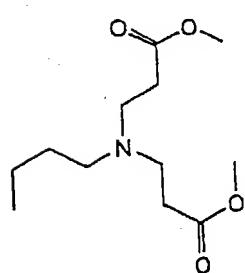
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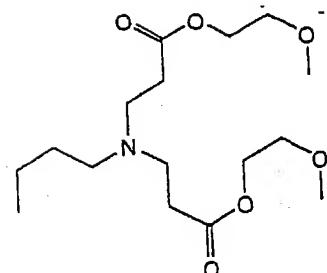
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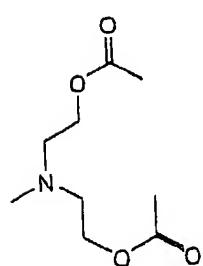
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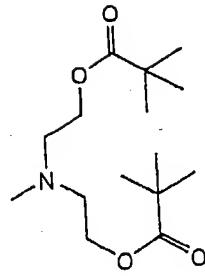
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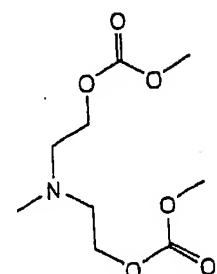
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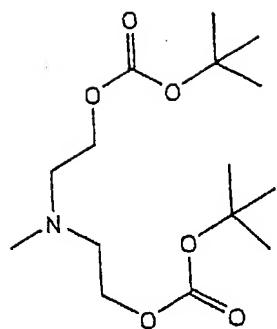
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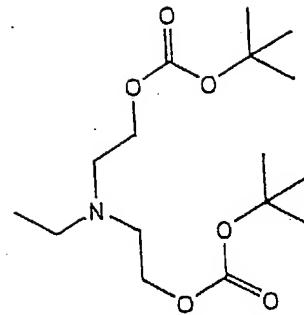
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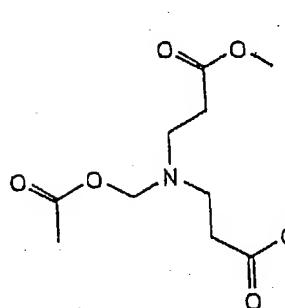
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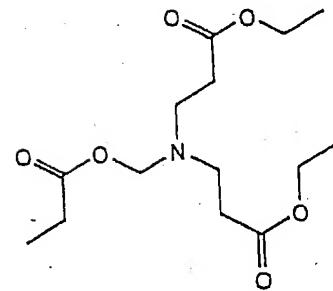
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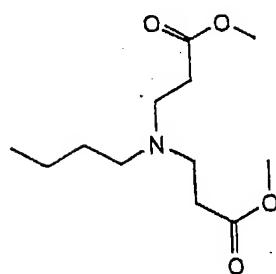
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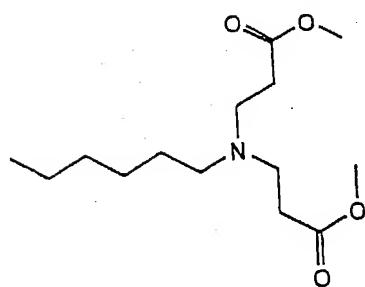
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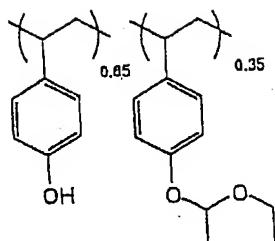
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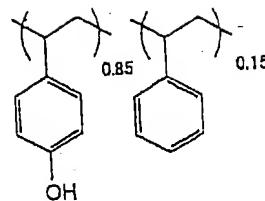
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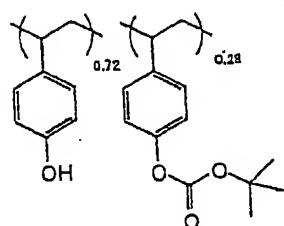
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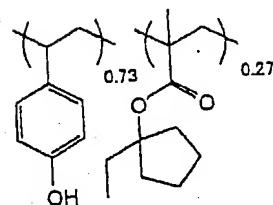
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Mw/Mn 1.10



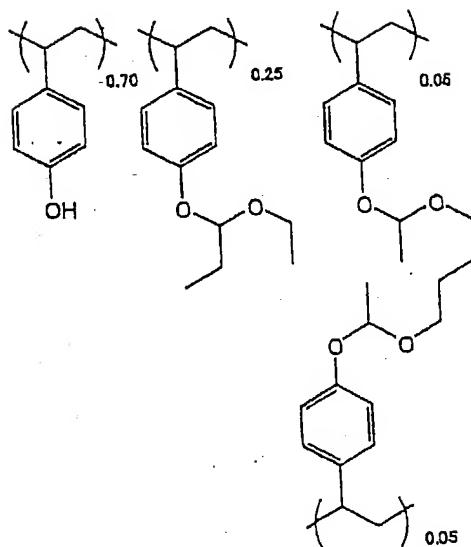
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Mw/Mn 1.10



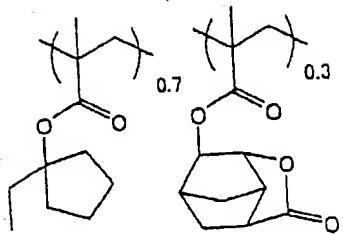
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Mw 10,000  
Mw/Mn 1.10



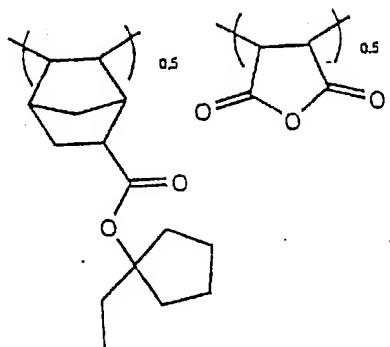
Polymer5  
Mw 12,000  
Mw/Mn 1.60



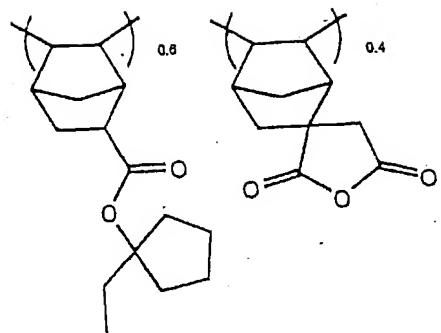
Polymer3  
Mw 16,000  
Mw/Mn 1.60



Polymer6  
Mw 10,000  
Mw/Mn 1.60



Polymer 7  
Mw 10,000  
Mw/Mn 1.50



Polymer 8  
Mw 18,000  
Mw/Mn 2.0

